Molecular Aspects of Antitumor Effects of a New Platinum(IV) Drug

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ABSTRACT

The new platinum(IV) complex cis,trans,cis-[PtCl2(CH3COO)2]-(NH3)(1-adamantylamine) [adamplatin(IV)] seems promising for the perspective application in therapy of corresponding tumors. It is therefore of great interest to understand details of mechanisms underlying its biological efficacy. Cellular uptake of the drug, alterations in the target DNA induced by platinum drugs along with processing of platinum-induced damage to DNA and drug inactivation by sulfur-containing compounds belong to major pharmacological factors affecting antitumor effects of platinum compounds. We examined in the present work the significance of these factors in the mechanism of antitumor effects of adamplatin(IV) and compared the results with those of the parallel studies performed with "conventional" cisplatin. The results show that deactivation of adamplatin(IV) by sulfur-containing compounds (such as glutathione or metallothioneins) is likely to play a less significant role in the mechanism of resistance of tumor cells to adamplatin(IV) in contrast to the role of these reactions in the effects of cisplatin. Moreover, the treatment of tumor cells with adamplatin(IV) does not result in DNA modifications that would be markedly different from those produced by cisplatin. In contrast, the effects of other factors, such as enhanced accumulation of the drug in cells, strong inhibition of DNA polymerization by these adducts, lowered DNA repair, and DNA-protein cross-linking are different from the effects of these factors in the mechanism underlying activity of cisplatin. Hence, the differences between effects of adamplatin(IV) and cisplatin observed in the present work on molecular level may help understand the unique activity of adamplatin(IV).
tential antitumor agents. Modifications include replacement of one or two nonleaving ammine groups with various organic carrier ligands and chloride groups with various leaving ligands. However, most of these modifications did not lead to the synthesis of the new compound that would have entered human clinical trials. Since the introduction of cisplatin, only cis-diamminecyclobutanedicarboxylatoplatinum(II) (carboplatin) and [(IR,2R-diamminecyclohexane)oxalato platinum(II)] (oxaloplatin) received worldwide approval and achieved routine clinical use. Carboplatin is less toxic than cisplatin and can be given at a much higher dose than cisplatin. Unfortunately, carboplatin is still only active in the same range of tumors as cisplatin. As of yet, oxaliplatin has not demonstrated substantial advantages over cisplatin or carboplatin. Oxaloplatin has shown potential for use in some cisplatin-resistant tumors if administered in combination with 5-fluorouracil and folinic acid, although combination of 5-fluorouracil with cisplatin also results in new antitumor effects in comparison with the treatment with single agents separately (Shiozaki et al., 2005).

The mode of action of platinum agents can be also considerably affected by other means than by changing of chemical structure. For example, originally inactive compounds can be activated only after they are delivered to the target tumor cells or to their intracellular components. Platinum(IV) complexes represent one example of this class of antitumor agents. Such compounds are frequently designated as prodrugs that have to be first, after their administration, activated by a reaction with reducing agents present in body liquids. Thus, the platinum(IV) complexes display potential advantages over their platinum(II) counterparts because of their greater stability and bioreductive activation, thereby allowing a greater proportion of the drug to arrive at the target intact. In addition, being inert to substitution, platinum(IV) complexes theoretically have the advantage of demonstrating fewer side effects and reduced drug loss owing to deactivation than their platinum(II) counterparts.

Until recently, the leading platinum(IV) compound designed as orally active antitumor drug was JM216 (Kelland, 2000). Two years ago, synthesis and cytotoxicity of a series of other platinum(IV) complexes with bulky hydrophobic ligands were reported (Turanek et al., 2004; Kozubik et al., 2005). The complex cis,trans,cis-[PtCl2(CH3COO)2(NH3)2](1-adamantylamine) [adamaplatin(IV)] (Fig. 1A) exhibited a high cytotoxic effect without cross-resistance to cisplatin in vitro assays, and the mode of synthesis was acceptable with respect to industrial manufacturing. The evaluation of cytotoxicity of adamaplatin(IV) on a panel of cisplatin resistant cancer cell lines revealed a high cytotoxic effect of this drug against leukemic, melanoma, and colorectal cancer cell lines and its high efficiency to trigger apoptosis (Turanek et al., 2004; Kozubik et al., 2005). The cytotoxic effect of adamaplatin(IV) on cisplatin-resistant cell lines is rapid and stronger in comparison with cisplatin and JM216 (Turanek et al., 2004). Thus, adamaplatin(IV) seems promising for the prospective application in therapy of corresponding tumors, so it is of great interest to understand details of molecular and biochemical mechanisms underlying the biological efficacy of adamaplatin(IV). Cellular uptake of the drug, alterations in the target DNA induced by platinum drugs along with processing of platinum-induced damage to DNA and drug inactivation by sulfur-containing compounds belong among major factors affecting antitumor effects of platinum compounds. Therefore, we examined the significance of these pharmacological factors in the mechanism of antitumor effects of adamaplatin(IV) and compared the results with those of the parallel studies performed with conventional cisplatin.

Materials and Methods

Chemicals. Cisplatin was obtained from Sigma (Prague, Czech Republic). Adamaplatin(IV) and cis-[PtCl2(NH3)2(1-adamantylamine)] [adamaplatin(II)] (Fig. 1A) were obtained from Prof. Giovanni Natile (University of Bari, Bari, Italy) and were prepared in the following way. K[PtCl2(NH3)] was prepared by a slight modification of the method of Abrams et al. (Intini et al., 2004). The cis-[PtCl2(NH3)(C10H15NH2)] [adamaplatin(II)] (K[PtCl2(NH3)] (0.30 g; 0.82 mmol) and tetraethylammonium chloride (0.21 g; 1.23 mmol) were mixed in dichloromethane (25 ml), and the suspension was stirred for 30 min. Solid KCl was removed by filtration of the orange solution. This solution was treated with 1-adamantylamine (0.12 g; 0.82 mmol) dissolved in 5 ml of dichloromethane and stirred at room temperature for 20 h. The yellow solution, taken to dryness by evaporation of the solvent under reduced pressure, afforded a yellow oily residue. Trituration with 20 ml of ethanol afforded a light yellow solid, which was separated by filtration of the mother liquor; washed with water, ethanol, and diethyl ether; and dried. The product proved to be the desired compound (yield 60%). Anal. Calcld. for C13H26Cl3KN2Pt, 414.067. Found: C, 26.34; H, 4.18; N, 3.97.

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\text{[PtCl2(NH3)(C10H15NH2)] [adamaplatin(II)]: K[PtCl2(NH3)] (0.30 g; 0.82 mmol) dissolved in 5 ml of dichloromethane and stirred at room temperature for 20 h. The yellow solution, taken to dryness by evaporation of the solvent under reduced pressure, afforded a yellow oily residue. Trituration with 20 ml of ethanol afforded a light yellow solid, which was separated by filtration of the mother liquor; washed with water, ethanol, and diethyl ether; and dried. The product proved to be the desired compound (yield 60%). Anal. Calcld. for C13H26Cl3KN2Pt, 414.067. Found: C, 26.34; H, 4.18; N, 3.97.}
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Adamaplatin(IV) was prepared from cis-[PtCl2(NH3)2(C10H15NH2)] according to the procedure previously published (Giandomenico et al., 1995). Anal. Calcld. for C13H26Cl3KN2Pt, 414.067. Found: C, 26.26; H, 4.18; N, 4.46. 1H NMR (CD3OD), δ (ppm; downfield from tetramethylsilane): 4.79 (2H, NH2), 3.89 (3H, NH3), 1.97 (3H, CH), 1.77 (6H, CH3), 1.53 (6H, AB, m, CH2). ESI mass spectrometry, m/z: 456 [M + Na]+. Adamaplatin(IV) was prepared from cis-[PtCl2(NH3)2(C10H15NH2)] according to the procedure previously published (Giandomenico et al., 1995). Anal. Calcld. for C13H26Cl3KN2Pt, 414.067. Found: C, 26.26; H, 4.18; N, 4.46. 1H NMR (CD3OD), δ (ppm; downfield from tetramethylsilane): 2.10 (3H, CH3), 2.06 (6H, OCH3), 2.05 (6H, CH3), 1.71 (6H, AB, m, CH2). ESI mass spectrometry, m/z: 574 [M + Na]+. Elemental analyses were performed using an Elemental Analyzer 1106 Carlo Erba instrument. 1H NMR spectra were recorded with a Bruker Avance DPX 300-MHz instrument (Bruker, New York, DE) operating at 300.13 MHz. ESI mass spectra were obtained by direct injection (10 μl/min) of 1 mM methanol solutions on an Agilent 1100 Series LC-MSD Trap System VL instrument (Agilent Technologies, Palo Alto, CA). Other
common reagents used to synthesize adamplatin(II) and adami- 
platin(IV) were from Sigma.

The stock solutions of the platinum compounds were prepared at the concentration of 1 × 10^2 M in 10 mM NaClO₄. They were stored at 4°C in the dark, or for the experiments with cells, in water.

Calf thymus (CT) DNA (42% G + C; mean molecular mass 20,000 
kJ/mol) was also prepared and characterized as described previously (Brabec and Palecek, 1976). Plasmids pSP73 (2464 bp), pSP73KB 
(2455 bp), pUC19 (2686 bp), and pBR322 (4363 bp) were isolated according to standard procedures. The synthetic oligodeoxyribonucleotides (Fig. 1B) were purchased from VBC-Genomics (Vienna, Austria) and purified as described previously (Brabec et al., 1999; Kasparkova et al., 2000). In the present work, their molar concentra- 
tions were related to the whole duplexes. The duplexes containing central sequence TTGTT/ACCA uniquely and site-specifically intra- strand cross-linked by adamiplatin(II) or cisplatin were prepared and characterized as described previously (Malina et al., 2002). Expression and purification of the rat HMGB1 domain A (HMGB1a; resi- 
dues 1–84) and HMGB1 domain B (HMGB1b; residues 85–180) were carried out as described previously (Stros, 2001). Restriction endo-
ucleases, T4 polynucleotide kinase, and Klenow fragment from 
DNA polymerase I (exonuclease minus, mutated to remove the 3′ → 5′ 
proofreading domain KF’') were purchased from New England Bio-
labs (Beverly, MA). Reverse transcriptase from human immunodefi-
ciency virus type 1 (RT HIV-1) was from Calbiochem (San Diego, 
CA). Riboprobe Gemini System II for transcription mapping contain-
ing T7 RNA polymerase was purchased from Promega (Madison, 
WI). Cell-free extracts (CFEs) were prepared from the HeLa S3 and 
CHO AA8 cell lines as described previously (Reardon et al., 1999). 
These extracts were kindly provided by T. Reardon and A. Sancar 
(University of North Carolina, Chapel Hill, NC). NF-κB (p50 dimer) 
was purchased from Active Motif (Rixensart, Belgium). GSH and 
ascorbic acid were purchased from Sigma, and rabbit MT-2 (Zn₇-
MT-2) was a kind gift of Prof. M. Vasak (University of Zurich, Zurich, 
Switzerland). Acrylamide, bis(acrylamide), urea, thioisourea, and 
NaCN were from Merck (Darmstadt, Germany). SDS was from Serva 
(Heidelberg, Germany). [γ-32P]ATP was from GE Healthcare (Little 
Chalfont, Buckinghamshire, UK). Proteinase K was from Roche Di-
agnostics (Mannheim, Germany).

Measurements of Platinum Accumulation in A2780 and 
A2780cisR Cells. Ovarian carcinoma cell lines A2780 and 
A2780cisR were cultivated and treated as described in a previously 
published article (Kasparkova et al., 2003b). The cells were treated with cisplatin or adamiplatin(IV) for 72 h at the concentrations cor-
responding to their IC₅₀ values in the cells tested (Kozubik et al., 
2005), i.e., A2780 or A2780cisR cells were treated with cisplatin at the concentration of 1.34 or 22.23 μM, respectively, and with adami-
platin(IV) at the concentration of 0.22 or 1.40 μM, respectively. The 
cells were analyzed for platinum by flameless atomic absorption 
spectrophotometry (FAAS). Standards and blank were prepared in the same way as the samples.

Transcription Mapping of DNA Adducts. Transcription of the 
(NdeI/HpaI) restriction fragment of pSP73KB DNA with DNA-de-
pendent T7 RNA polymerase and electrophoretic analysis of trans-
scripts were performed according to the protocols recommended by 
and previously described in detail in Brabec and Leng (1993).

DNA Interstrand Cross-Linking. The amounts of interstrand 
cross-links (CLs) formed by adamiplatin(II) in linear DNA were mea-
ured in pSP73 plasmid (2464 bp), which was first linearized by 
EcoRI (EcoRI cuts only once within pSP73 plasmid) and subse-
sequently modified by adamiplatin(II). The samples were analyzed for the interstrand CLs by agarose gel electrophoresis under denaturing 
conditions. Upon electrophoresis under denaturing conditions, 3′ end-labeled strands of linearized pSP73 plasmid containing no in-
terstrand CLs migrate as a 2464-base single strand, whereas the 
interstrand cross-linked strands migrate more slowly as a higher 
molecular mass species. The radioactivity associated with the indi-
vidual bands in each lane was measured to obtain estimates of the 
fraction of non–cross-linked or cross-linked DNA under each condi-
tion. The frequency of interstrand CLs [the amount of interstrand 
CLs per one molecule of adamiplatin(II) bound to DNA] was calcu-
lated using the Poisson distribution from the fraction of non–cross-
linked DNA in combination with the rᵢ values and the fragment size. Other details of this assay can be found in previously published 
articles (Brabec and Leng, 1993; Brabec et al., 1999).

GeL Mobility Shift Assay Reactions with HMG Domain Pro-
teins. The 5′ end-labeled 20- or oligonucleotide duplexes (TGTT 
shown in Fig. 1B) either unmodified (controls) or containing the 
central platinum intrastrand CL in their top strands, were used, and 
their reaction with HMG domain proteins was performed and ana-
yzed as described previously (He et al., 2000). The values of appar-
ent dissociation constant, Kₐpp, were estimated in the manner 
described in He et al. (2000). Each Kₐpp is the average of at least 
two measurements.

Inhibition of DNA Polymerization. We investigated in the 
present work DNA polymerization using the templates site-spe-
cifically modified by adamiplatin(II) or cisplatin by DNA polymerase, 
namely, RT HIV-1. The 30-mer templates containing a single 1,2-GG 
intrastrand CL of adamiplatin(II) or cisplatin were prepared in the same way as described in our recent work (Novakova et al., 2003, 
2005). 17-mer DNA primer was complementary to the 3′ termini of 
the 30-mer templates, respectively. The DNA substrates were 
formed by annealing templates and 5′ end-labeled primers at a 
molar ratio of 3:1. Experiments were performed at 37°C in a volume 
of 50 μl in a buffer containing 50 mM Tris.HCl, pH 7.4, 10 mM 
MgCl₂, 0.1 mM dithiothreitol, 50 μg/ml bovine serum albumin, 0.1% 
Nonidet P-30, 100 μM dATP, 100 μM dCTP, 100 μM dGTP, 100 μM 
TTP, and 1.0 U of RT HIV-1. Reactions were terminated by the 
addition of EDTA, so that its resulting concentration was 20 μM 
and by heating at 100°C for 30 s. Products were resolved by denaturing 
24 or 15% PAA/8 M urea gel. Other details of these experiments can 
be found in our previously published articles (Novakova et al., 2003, 
2005).

Repair Synthesis by Human Cell Extracts. Repair DNA syn-
thesis of CFEs was assayed using pUC19 and pBR322 plasmids. 
Each reaction of 50 μl contained 250 ng of nonmodified pBR322 
and 250 ng of modified or platinated pUC19, 2 mM ATP, 30 mM KCI, 
0.5 mg/ml creatine phosphokinase (rabbit muscle), 20 mM each 
dGMP, dCTP, and TTP, 8 mM dATP, 74 kb of [α-32P]dAMP in the 
buffer composed of 40 mM HEPES-KOH, pH 7.5, 5 mM MgCl₂, 
0.5 mM dithiothreitol, 22 mM creatine phosphate, 1.4 mg/ml bovine 
serum albumin, and 150 μg of CFE from the HeLa S3 cells. Reactions 
were incubated for 3 h at 25°C and terminated by adding EDTA to 
the final concentration of 20 mM, SDS to 0.6%, and proteinase K to 250 
μg/ml and then incubating for 30 min. The products were extracted 
with 1 volume phenol/chloroform (1:1). The DNA was precipitated 
from the aqueous layer by the addition of 1/50 volume of 5 M NaCl, 
5 mg of glycerin, and 2.5 volumes of ethanol. After 20 min of 
icubation on dry ice and centrifugation at 12,000g for 30 min at 4°C, 
the pellet was washed with 0.5 ml of 70% ethanol and dried in a 
vacuum centrifuge. DNA was finally linearized before electrophore-
sis on a 1% agarose gel containing 0.3 mg/ml ethidium bromide.

Nucleotide Excision Assay. The 149-bp substrates containing a 
central, single 1,2-GG intrastrand CL of adamiplatin(II) or cisplatin 
were assembled from three oligonucleotide duplexes as described in 
our previous work (Malina et al., 2002). The central duplex was 
TGTT(NER) duplex (Fig. 1B) to which two duplexes (arms) with 
random base pair sequences with overhangs partially overlapping 
those of the modified duplex were ligated (one to each side) by T4 
DNA ligase. The top strand of the modified central duplexes were 5′ 
end labeled with 32P before ligation. Full-length substrates [non-
modified, containing the 1,2-intrastrand CL of adamiplatin(II) or 
cisplatin] were generated from unligated products on a denaturing 
6% PAA gel, purified by electroelution, reannealed, and stored in 
annealing buffer (50 mM Tris-HCl, pH 7.9, 100 mM NaCl, 10 mM
MgCl₂ and 1 mM dithiothreitol) at 20°C. In vitro repair of the 1,2-GG intrastrand CL of cisplatin or adamplatin(II) was measured in an excision assay as described previously (Malina et al., 2002) with minor modifications. The reaction mixtures (25 μl) contained 10 fmol of radiolabeled DNA, 50 μg of CFE from the HeLa S3 or CHO AA8 cell lines, 20 μM dATP, 20 μM dGTP, 20 μM dCTP, and 20 μM dTTP in reaction buffer (23 mM HEPES, pH 7.5, 4 mM KCl, 4.8 mM MgCl₂, 0.16 mM EDTA, 0.52 mM dithiothreitol, 1.5 mM ATP, 5 μg of bovine serum albumin, and 2.5% glycerol) and were incubated at 30°C for 40 min. DNA was deproteinized and precipitated by ethanol. Reaction products were treated overnight with 0.4 M NaCN, pH 10 to 11, at 45°C and precipitated by ethanol before resolution on the gels. The excision products were separated on denaturing 10% PAGE gels. The excision products were treated overnight with 0.4 M NaCN, pH 10 to 11, at 45°C and precipitated by ethanol before resolution on the gels. The excision products were separated on denaturing 10% PAGE gels.

Reactions with Ascorbic Acid. The kinetic measurements of reactions of ascorbic acid with adamplatin(IV) were investigated by conventional absorption spectrophotometry monitoring a decrease of absorption in the range of 300 to 400 nm of solutions containing the platinum complex and ascorbic acid exactly as described in the previous work (Levina et al., 2000). The platinum compound was mixed with ascorbic acid at 37°C in the medium of 0.04 M Tris-HClO₄ buffer, pH 7.0, in the dark. Reactions were initiated by mixing the platinum complex with the buffer followed by immediate addition of GSH or MT-2.

Physical Methods. Absorption spectra were measured with a Beckman DU-7400 spectrophotometer (Beckman Coulter, Fullerton, CA). FAAS measurements were carried out with a Varian AA240Z Zeeman atomic absorption spectrometer (Varian, Inc., Palo Alto, CA) equipped with a GTA 120 graphite tube atomizer. For FAAS analysis, the cells were digested in nitric acid followed by the addition of H₂O₂ and HCl, whereas DNA was precipitated with ethanol and dissolved in 0.1 M HCl. Differential pulse polarography curves were recorded with the aid of a PAR electrochemical analyzer (model 384B; EG&G Berthold, Bad Wildbad, Germany). Purification of oligonucleotides with the aid of high-performance liquid chromatography was carried out on a Waters high-performance liquid chromatography system consisting of Waters 2690 pump, Waters 2487 UV detector, and Waters 600S controller with MonoQ HR 5/5 column (Waters, Milford, MA). The gels were visualized by using the BAS 2500 FUJIFILM bioimaging analyzer (Fujifilm, Tokyo, Japan), and the radioactivities associated with bands were quantitated with the AIDA image analyzer software (Raytest, Straubenhardt, Germany).

Results

Cellular Platinum Complex Uptake. A possible explanation for the difference in cytotoxicity between cisplatin and the complexes containing alkylamine ligands with increased hydrophobicity, such as JM216 or adamplatin(IV) may be associated with differential cellular uptake of the compounds (Turanek et al., 2004). Cellular uptake of cisplatin and adamplatin(IV) was measured in A2780 and A2780cisR cells (sensitive and resistant to cisplatin, respectively) treated with the platinum compound for 72 h at the concentrations corresponding to the IC₅₀ values in the cells tested. These cancer cell lines sensitive and resistant to cisplatin, respectively, were among those used for testing activity of adamplatin(IV) in which this drug exhibited a markedly enhanced activity in comparison with cisplatin (IC₅₀ values found for cisplatin and adamplatin(IV) in A2780 cells were 1.34 and 0.22 μM, respectively, whereas these values found for cisplatin and adamplatin(IV) in A2780cisR cells were 24.23 and 1.40 μM, respectively (Kozubík et al., 2005). The results show that platinum associated with the cells treated with adamplatin(IV) was considerably higher than that with cisplatin (20 or 30 times in the sensitive and resistant cells, respectively; Table 1). Thus, the enhanced cellular uptake of adamplatin(IV) is accompanied by an increased cellular uptake of the platinum complex with the buffer followed by immediate addition of GSH or MT-2.

**Table 1**

<table>
<thead>
<tr>
<th>Platinum uptake in cells</th>
<th>A2780</th>
<th>A2780cisR</th>
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<tbody>
<tr>
<td>Cisplatin</td>
<td>21 ± 2</td>
<td>3.7 ± 0.4</td>
</tr>
<tr>
<td>Adamplatin(IV)</td>
<td>422 ± 10</td>
<td>110 ± 5</td>
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</table>

Table shows the uptake of cisplatin and adamplatin(IV) into A2780 and A2780cisR after 72 h. Each value (in picomoles of platinum per 10⁶ cells per microliter) represents the average of six samples ± S.E.
tin(IV) seems to be an important factor responsible for enhanced cytotoxicity of this platinum complex in comparison with conventional cisplatin. In contrast, the uptake of adamplatin(IV) in A2780 cells was ca. 4 times higher than in A2780cisR cells, whereas the toxicity of adamplatin(IV) in A2780 cells was ca. 6 times higher than that in A2780cisR cells. Hence, the cytotoxicity does not simply match the levels of cellular uptake, so that it is intuitively appealing to suggest that the cytotoxicity of adamplatin(IV) may also be related to other pharmacological factors affecting platinum drug cytotoxicity.

**DNA Modifications in Cell-Free Media.** Platinum-DNA adducts induce a cellular signaling response regulating survival and apoptosis or necrosis. Platinum(IV) complexes undergo ligand substitution reactions that are slow relative to their platinum(II) counterparts. The antitumor activity of platinum(IV) complexes is therefore likely to require in vivo reduction to the kinetically more labile, and therefore reactive, platinum(II) congeners. Indeed, previous studies have shown that platinum(IV) metal centers are readily reduced by cellular components, such as small molecules involved in redox balance in the body (Hall and Hambley, 2002) to form the platinum(II) analogs that bind more rapidly to DNA. Thus, platinum(IV) compounds can be reduced extracellularly and enter the tumor cell as platinum(II) congener, or intracellularly having entered as platinum(IV) compound (Hall and Hambley, 2002). It implies that the administered adamplatin(IV) serves as a prodrug for the chemotherapeutically active platinum(II) analog adamplatin(II) (Fig. 1B), which is presumably the complex that binds to DNA (Brabec, 2002; Hall and Hambley, 2002). In the present study, we have applied some methodologies previously developed for cisplatin and its analogs to investigate the reaction products of adamplatin(II) with DNA. Because this complex may be interesting from a clinical point of view, it is important to determine the extent to which it resembles the parent drug cisplatin in binding to its biological target(s).

Solutions of CT DNA at a concentration of 0.32 mg/ml were incubated with adamplatin(II) at the $r_f$ value in the range of 0.005 to 0.1 in 10 mM NaClO$_4$ at 37°C. $r_f$ is defined as the molar ratio of free platinum complex to nucleotide phosphates at the onset of incubation with DNA. At various time intervals, an aliquot of the reaction mixture was withdrawn and assayed by differential pulse polarography for platinum not bound to DNA. The amount of platinum bound to DNA ($r_f$) was calculated by subtracting the amount of free (unbound) platinum from the total amount of platinum present in the reaction. The amount of platinum coordinated to DNA increased with time with a rate identical to that found for cisplatin, and after approximately 24 h, adamplatin(II) was quantitatively bound. It is noteworthy that the analytical method used monitors the covalent attachment of platinum complexes to DNA (so that the results are not affected by the subsequent closure of monofunctional adducts to bifunctional lesions) (Kim et al., 1990). pH of the reaction mixture containing DNA and adamplatin was measured within 48 h after mixing DNA with the platinum complex, and no changes in pH were noticed.

Unless stated otherwise, the samples of DNA modified by adamplatin(II) and cisplatin complexes, which were further analyzed in this work by biochemical or biophysical techniques, were prepared by incubating DNA with the platinum complexes for 48 h (also see Materials and Methods). The binding experiments indicate that such platination reactions resulted in the coordination of all molecules of the platinum complexes, which made it possible to prepare easily and precisely the samples of DNA modified by the platinum complexes at a preselected value of $r_f$. It was also verified that the analyses of DNA samples modified by adamplatin(II) by using biochemical and biophysical techniques described in this work yielded identical results if aged or fresh solutions of platinum complexes were used.

In vitro RNA synthesis by RNA polymerases on DNA templates containing several types of bifunctional adducts of platinum complexes can be prematurely terminated at the level or in the proximity of adducts (Brabec and Leng, 1993). Cutting of pSP73KB DNA (Brabec and Leng, 1993) by NdeI and HpaI restriction endonucleases yielded a 212-bp fragment (a substantial part of its nucleotide sequence is shown in Fig. 2B). This fragment contained T7 RNA polymerase promoter (in the upper strand close to its 3′ end; Fig. 2B). The...
experiments were carried out using this linear DNA fragment, modified by cisplatin and adamplatin(II) at \( r_n = 0.005 \), for RNA synthesis by T7 RNA polymerase (Fig. 2, lanes cisPt and adamPt). RNA synthesis on the template modified by the platinum complexes yielded fragments of defined sizes, which indicates that RNA synthesis on these templates was prematurely terminated (Fig. 2A). The sequence analysis revealed that the major bands resulting from termination of RNA synthesis by the adducts of cisplatin and adamplatin(II) were identical and occurred at guanine (G) sites and to a considerably lesser extent at adenine (A) sites. These G and A sites were mostly contained in GG or AG sites, which are preferential DNA binding sites for conventional cisplatin. Taken together, the results of the transcription mapping experiments suggest that base sequence selectivity of cisplatin and adamplatin(II) are similar and that also the major adducts formed on DNA by adamplatin(II) and cisplatin are identical, presumably 1,2-GG or AG intrastrand CLs.

Cisplatin also forms in DNA a small amount of interstrand CLs. Despite their low abundance, these lesions are also considered relevant to the antitumor effects of cisplatin (Brabec, 2000). The amounts of interstrand CLs formed by adamplatin(II) in linear DNA were measured in linearized pSP73 plasmid, which was modified by adamplatin(II) to various values of \( r_n \). The samples were analyzed for the interstrand CLs by agarose gel electrophoresis under denaturing conditions (Brabec and Leng, 1993; Brabec et al., 1999). The bands corresponding to more slowly migrating interstrand cross-linked fragments were noticed if adamplatin(II) was used to modify linearized DNA at \( r_n \) as low as \( 5 \times 10^{-4} \) (Fig. 3). The intensity of the more slowly migrating band increased with the growing level of the modification. The radioactivity associated with the individual bands in each lane was measured to obtain estimates of the fraction of non–cross-linked or cross-linked DNA under each condition. The frequency of interstrand CLs produced by adamplatin(II) was somewhat lower (~4%) than cross-linking efficiency of cisplatin (6%; Brabec and Leng, 1993).

**Recognition of Major 1,2-GG Intrastrand DNA Adducts by HMG-Domain Proteins, Translesion DNA Synthesis across These Adducts, and Their Repair.** An important feature of the mechanism of action of cisplatin is that the major adducts of this platinum drug (1,2-GG and AG intrastrand CLs) are recognized by proteins containing HMG domains (Brabec, 2002). It has also been shown (Brabec, 2002) that the binding of these proteins to cisplatinated DNA mediates the antitumor effects of this platinum drug. In addition, the replacement of one or both ammine groups in cisplatin by various, mostly cyclic nonleaving ligands, markedly affected affinity of HMG-box proteins to the 1,2-GG intrastrand CLs (Wei et al., 2001). Hence, it was of great interest also to examine whether the replacement of the ammine group in cisplatin by 1-adamantylamine could also affect affinity of HMG-box proteins to this major adduct.

The interactions of the rat HMGB1a (residues 1–84) and HMGB1b (residues 85–180) with the 1,2-GG intrastrand CLs of adamplatin(II) were investigated using gel mobility shift assay. In these experiments, the 20-bp duplex TGGT (Fig. 1B) was modified so that it contained a single, site-specific 1,2-GG intrastrand CL of adamplatin(II) or cisplatin. Because adamplatin(II) contains two different nonleaving ligands, two orientational isomers (5′ and 3′) could be formed similarly as in the case of 1,2-GG intrastrand CL of cisplatin or adamplatin(II). However, adamplatin(II) contains two different nonleaving ligands, two orientational isomers (5′ and 3′) could be formed similarly as in the case of 1,2-GG intrastrand CL of cisplatin or adamplatin(II).

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ence of a shifted band whose intensity increases with increasing protein concentration, both HMGB1a and HMGB1b recognize the duplex containing the 1,2-GG intrastrand CL of adamplatin(II) (Fig. 4). These data indicate that HMGB1a binds the probe containing the 1,2-GG intrastrand CL of adamplatin(II) with a relatively high affinity, although somewhat lower than to the probe containing the same adduct of cisplatin (Fig. 4A). Evaluations of these titration data afforded $K_{\text{app}}$ values of $4.0 \times 10^{-8}$ and $5.2 \times 10^{-8}$ M for the binding of HMGB1a to the adducts of cisplatin and adamplatin(II), respectively. The titration of the 20-bp duplex containing the 1,2-GG intrastrand CL of adamplatin(II) with HMGB1b has revealed (Fig. 4B) that this protein also binds to the 1,2-GG intrastrand CL of adamplatin(II) with a similar affinity as to the probe containing 1,2-GG intrastrand CL of cisplatin. Evaluations of these titration data afforded $K_{\text{app}}$ values of $3.8 \times 10^{-7}$ and $4.0 \times 10^{-7}$ M for the binding of HMGB1a to the adducts of cisplatin and adamplatin(II), respectively. The affinity of HMGB1b to these platinum lesions was, however, considerably smaller than that of HMGB1a as expected (Wei et al., 2001; Malina et al., 2002). Hence, it is reasonable to expect that the full-length HMGB1 protein, which is an abundant chromosomal protein in mammalian cells and the candidate protein for participation in the repair shielding mechanism of antitumor activity of cisplatin (Zamble and Lippard, 1999; Brabec, 2002), has a similar affinity to the 1,2-GG intrastrand CL of adamplatin(II).

DNA secondary structures have significant effects on processivity of a number of prokaryotic, eukaryotic, and viral DNA polymerases. The character of DNA templates containing site-specific platinum adducts dictates whether DNA polymerases are blocked or can traverse through the lesion. DNA polymerization using the templates site-specifically modified by 1,2-GG intrastrand CLs of adamplatin(II) or cisplatin by RT HIV-1 was investigated to reveal the potential differences imposed on DNA by these two adducts. We constructed the 17-mer/30-mer primer-template duplexes TGGT (Fig. 5A) unplatinated or containing the 1,2-GG intrastrand CL of either cisplatin or adamplatin(II). The first 17 nucleotides on the 3’ terminus of the 30-mer template strand were complementary to the nucleotides of the 17-mer primer, and the 3’ guanine involved in the 1,2-GG CL of cisplatin or adamplatin(II) on the template strand was located at its 20th position from the 3’ terminus (Fig. 5A). After annealing the 17-nucleotide primer to the 3’ terminus of the unplatinated or platinated template strand (positioning the 3’ end of the primer three bases before the adduct in the template strand), we examined DNA polymerization through the unique CLs on the template by RT HIV-1 in the presence of all four deoxyribonucleoside 5’-triphosphates. The reaction was stopped at various time intervals, and the products were analyzed using a sequencing gel (Fig. 5A). Polymerization using the template containing the CL of cisplatin proceeded rapidly up to the nucleotides at the sites opposite the CL, such that the 20- and 21-nucleotide products accumulated to a significant extent (Fig. 5A, lanes 6–10). There was no accumulation of DNA intermediate products with the 30-mer control template as the full-length product was being formed (Fig. 5A, lanes 1–5). The full-length products were also noticed with the 30-mer template containing the CL of cisplatin, although in a smaller amount. The results are in agreement with previously published work (Marini et al., 2005) and confirm that the 1,2-GG intrastrand CL of cisplatin inhibits DNA synthesis, but translesion synthesis may occur. In contrast, under the same experimental conditions, DNA polymerization by RT HIV-1 on the template containing 1,2-GG intrastrand CL of adamplatin(II) (Fig. 5A, lanes 11–15) proceeded up to the site preceding the platinated GG sequence so that also the 19-nucleotide product accumulated to a significant extent and then also the 20-nucleotide product accumulated corresponding to the nucleotide at the site opposite the 5’ G involved in the CL. There was almost no accumulation of shorter and longer DNA intermediates, and importantly, fewer full-length products accumulated than when DNA polymerization using the template site-specifically modified by 1,2-GG intrastrand CLs of cisplatin was examined (Fig. 5B). This result indicates that the 1,2-GG intrastrand CLs of adamplatin(II) impede elongation of DNA to a greater extent than the major adducts of cisplatin (Fig. 5B).

The high degree of structural and sequence conservation of the domains among eukaryotic, prokaryotic, and viral polymerases suggests that the results from studies of the RT HIV-1 should also be applicable to other DNA polymerases. Hence, the stronger inhibition of DNA polymerization by the adamplatin(II) 1,2-GG intrastrand CL reflects an important

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**Fig. 5.** Primer extension activity of RT HIV-1. A, experiments were conducted using the 17-mer/30-mer primer/template duplex for various times using undamaged template (lanes 1–5), the template containing 1,2-GG intrastrand CL of cisplatin (lanes 6–10), or adamplatin(II) (lanes 11–15). Timings were as follows: 10 min, lanes 1, 6, and 11; 20 min, lanes 2, 7, and 12; 30 min, lanes 3, 8, and 13; 60 min, lanes 4, 9, and 14; and 90 min, lanes 5, 10, and 15. The pause sites opposite the platinated guanines are marked 19 and 20. The pause sites opposite the nucleotides 17 and 21 of the template are marked 17 and 21, respectively. The nucleotide sequence of the primer and the templates including its nucleotide numbering are shown beneath the gel. B, The time dependence of the inhibition of DNA synthesis on undamaged (control) template (squares), DNA containing 1,2-GG intrastrand CL of cisplatin (triangles), and DNA containing 1,2-GG intrastrand CL of adamplatin(II) (circles) is indicated. Data are means ± S.E. from three different experiments with two independent template preparations.
difference in biological processes of replication or DNA repair in comparison with the major adduct of cisplatin.

Figure 6 illustrates an experiment that measures DNA repair synthesis by a repair-proficient HeLa CFE in pUC19 plasmid modified at \( r_p = 0.05 \) by adamplatin(II) or cisplatin. Repair activity was monitored by measuring the amount of incorporated radiolabeled nucleotide. A similar amount of undamaged pBR322 of a slightly different size is included in the reactions to show the background incorporation into undamaged plasmid. This background incorporation was subtracted from that found for metallated pUC19 plasmid. Different levels of damage-induced DNA repair synthesis were detected in the plasmid modified by adamplatin(II) and cisplatin (Fig. 6B). The level of the synthesis detected in the plasmid modified by adamplatin(II) was by \(-25\%\) lower than that in the plasmid modified by cisplatin.

DNA repair synthesis can be due to various DNA repair mechanisms. Bulky, helix-distorting DNA adducts, such as those generated by various chemotherapeutics, including cisplatin, are removed from DNA by nucleotide excision repair (NER), which is an important component of the mechanism underlying biological effects of these agents. Efficient removal of CLs formed in DNA by platinum antitumor compounds has already been reported for various NER systems, including human and rodent excinuclease (Zehnulova et al., 2001; Kasparkova et al., 2003a). The result presented in Fig. 7A, lane 4, is consistent with these reports. The major excision fragment contains 28 nucleotides, and other primary excision fragments are 24–29 nucleotides (Readon et al., 1999; Zehnulova et al., 2001; Kasparkova et al., 2003a). The 1,2-GG intrastrand CL of adamplatin(II) was also excised by both human and rodent excinuclease (shown for rodent excinuclease in Fig. 7A, lane 6), although with a lower efficiency than the major intrastrand CL of cisplatin. Thus, these results demonstrate the lower efficiency (by ca. 20%) of the mammalian NER systems used in the present work to excise major intrastrand adducts of adamiplatin(II) (Fig. 7B).

**DNA-Protein Cross-Linking.** When we isolated and purified the short oligonucleotide duplexes containing single, site-specific 1,2-GG intrastrand adduct of adamiplatin(II) or cisplatin as DNA probe for the studies of affinity of HMG-domain proteins to this adduct (vide supra), we noticed that the intrastrand CL was formed by adamiplatin(II) noticeably more slowly, although the overall level of the oligonucleotide platination was similar. A reasonable explanation of this observation is that adamiplatin(II) forms on DNA more monofunctional adducts that close to bifunctional CL less readily than monofunctional adducts of cisplatin. Thus, the monofunctional adducts of adamiplatin(II) might persist for a considerably longer time than the same adducts of cisplatin. To test this hypothesis, CT DNA was incubated in 10 mM NaClO\(_4\) at 37°C for 24 h with adamiplatin(II) or cisplatin so that the level of platination corresponded to \( r_p \) values of 0.01, 0.05, and 0.1. We trapped the reactive monofunctional adducts by fixing radioactive (tritiated) thiourea on the platinum-DNA complex (Boudny et al., 1992; Brabec et al., 1996). The intriguing observation was that the radioactivity associated with DNA treated with adamiplatin(II) was more than 3 times higher than that with cisplatin, confirming that adamiplatin(II) forms reactive monofunctional adducts with pronouncedly higher frequency than cisplatin.

Adamiplatin(II) and cisplatin were investigated for their ability to form ternary DNA-protein complexes. The 23-bp duplex (DP) (the central part of its nucleotide sequence shown in Fig. 1B corresponded to DNA consensus sequence of NF-kB) 5′ end-labeled at its top strand was globally modified by adamiplatin(II) or cisplatin for 24 h to the \( r_p \) value that corresponded to 0.5 platinum atom bound per duplex. The duplex modified by adamiplatin(II) or 10 nM cisplatin was mixed with HMGB1a, KF\(^-\), or NF-kB (p50) dimers (the molar ratio protein duplex was 30, 1, and 5, respectively). The different molar ratios were used to reflect the different binding affinity \([K_{D_{app}}]\) of the proteins to the platinated duplex. Ternary DNA-platinum-protein cross-linking efficiency was assessed by SDS-PAGE shift assay. These proteins were chosen as representatives of DNA-binding proteins with structural, enzymatic, or regulation function. Fractions were detected by SDS-PAGE with significantly retarded mobility (Fig. 8, lanes 6–8 and 10–12) compared with that of the free probe (Fig. 8, lanes 1, 5, and 9). These more slowly migrating fractions were eliminated after treatment with NaCN or proteinase K converting them to those of the unmodified probes (data not shown). These results suggest that the species is a protein-DNA CL tethered by platinum-DNA and platinum-protein covalent bonds. Although the proteinase K and NaCN experiments clearly indicate that protein is the species cross-linked to DNA, the amino acids participating in the cross-linking reaction have not been determined. It is noteworthy that the amount of radioactivity associated with the bands corresponding to DNA-protein CLs formed by adamiplatin(II) was 6 to 7 times higher than that by cisplatin, demonstrating that adamiplatin(II) exhibits a considerably higher efficiency to form ternary DNA-platinum-protein CLs than cisplatin.

We have also determined the efficiency of adamiplatin(IV) and cisplatin to form DNA-protein CLs in A2780 and A2780cisR cells after their exposure in culture to these platinum drugs using a potassium-SDS assay (Zhitkovich and Costa, 1992). The cells were treated with adamiplatin(IV) or
cisplatin at the concentrations corresponding to the IC$_{50}$ values in the cells tested (vide supra) for 72 h. The DNA-protein cross-linking was evaluated using DPC (Zhitkovich and Costa, 1992). Because the cellular uptake of cisplatin and adamplatin(IV) is different (vide supra), the values of DPC were normalized to the identical amount of platinum associated with the cells. Exposure of A2780 and A2780cisR cells to adamplatin(IV) resulted in the formation of substantial amounts of DNA-protein CLs (Table 2). The efficiency of adamplatin(IV) in sensitive and resistant cells was 4 and 9 times higher, respectively, than that found for the exposure of these cells to cisplatin.

Reactions with Sulfur-Containing Compounds and Ascorbic Acid. Platinum(II) or platinum(IV) compounds have a strong thermodynamic preference for binding to sulfur donor ligands. Hence, before antitumor platinum drugs reach DNA in the nucleus of tumor cells, they may interact with various compounds, including sulfur-containing molecules. These interactions are generally thought to play a role in mechanisms underlying tumor resistance to platinum compounds, their inactivation, and their side effects (Reedijk, 1999). Therefore, interest in the interactions of platinum antitumor drugs with sulfur-containing molecules of biological significance has recently increased markedly.

Examples of endogenous thiols to which platinum complexes bind when they are administered intravenously or after they enter the cell are GSH and MT. MTs are small (30- to 70-kDa) cellular metalloproteins that strongly bind metal ions. The metal ion binding domains of MT-2 consist of 20 cysteine residues arranged in two thiol (S)-rich sites (Romero-Isart and Vasak, 2002). In the present work, we investigated reactions of GSH and mammalian MT-2 with adamplatin(IV), adamplatin(II), and cisplatin using UV-absorption spectrophotometry (Hagrman et al., 2003).

Adamplatin(IV), adamplatin(II), or cisplatin at a concentration of 33 μM was mixed with 5 mM GSH (this concentration of GSH is physiologically relevant; Romero-Isart and Vasak, 2002) at 37°C in the medium of 10 mM NaClO$_4$ plus 10 mM Tri-HCl buffer, pH 7.4. The half-times of the reactions, which mainly result in the formation of platinum-sulfur bonds, were 10 and 113 min for reactions of adamplatin(IV) and adamplatin(II) with GSH, respectively. Cisplatin reacted with GSH with the half-time of 11 min [i.e., with a similar rate to that of adamplatin(IV) but faster than adamplatin(II)].

Adamplatin(II) or cisplatin at the concentration of 200 μM were also mixed with MT-2 at the concentration of 2.1 μM under the same conditions. The half-times of the reaction of adamplatin(II) or cisplatin with MT-2 were similar (42 or 40 min, respectively). In contrast, no changes in the absorbance at 260 nm, indicating reaction of MT-2 with adamplatin(IV), were noticed if MT-2 at the concentration as high as 200 μM was incubated with 200 μM adamplatin(IV) for 1 week.

The anticancer activity of adamplatin(IV) is likely to be due to its reduction to more reactive platinum(II) analog in agreement with the assumption that these compounds act as prodrugs to their platinum(II) analogs. Hence, reduction of adamplatin(IV) by potential bioreductant such as ascorbic acid may be also important for understanding of the mechanism of antitumor activity of this platinum complex (Lemma et al., 2000). Adamplatin(IV) at the concentration of 0.15 mM was mixed with 1.5 mM ascorbic acid (this concentration of ascorbic acid is physiologically relevant; Lemma et al., 2000). The half-time of the reaction was found 15 min. Thus, reduction of adamplatin(IV) is fairly rapid similarly as reduction of JM216 by ascorbic acid (Lemma et al., 2000) compared with hydrolytic biotransformation pathways.
Mechanism of Antitumor Effects of New Pt(IV) Drug

Discussion

Adamplatin(IV) proved to be quite toxic to the tumor cells, with IC\text{50} values that were markedly lower than those observed for cisplatin and JM216 (Turanek et al., 2004; Koizub, et al., 2005). The origin of this effect may be multifactorial. The major pharmacological factors affecting platinum drug cytotoxicity are cellular uptake and efflux; structure; and frequency of target (DNA) adducts, including their intracellular processing; and metabolic deactivation by sulfur nucleophiles (Brabec and Kasparkova, 2005). All these factors combine to affect signaling pathways, leading to apoptosis or necrosis of tumor cells.

The present study reveals that the cellular uptake of adamplatin(IV) is considerably greater than that of cisplatin and JM216 (O’Neill et al., 1999). An increase of hydrophobicity of ligands of various platinum complexes has been shown to positively correlate with enhanced cellular uptake (Moeller et al., 2000). Thus, the observation that the cellular uptake of adamplatin(IV) is greater than that of the closely related JM216 is consistent with a considerably more hydrophobic character of alkylamine ligands in adamplatin(IV) (Turanek et al., 2004). The cellular accumulation was enhanced in A2780cisR cells resistant to cisplatin over sensitive A2780 cells, suggesting a mechanism for enhancement of tumor cell selectivity toward cisplatin-resistant cells. However, there is no clear correlation between uptake and cytotoxicity of adamplatin(IV) within the tumor cells sensitive and resistant to cisplatin. Nevertheless, the results confirm the remarkable uptake of adamplatin(IV) so that at least part of the increased cytotoxicity of adamplatin(IV) over cisplatin seems to be attributable to an increased intracellular accumulation.

It is widely held that reduction to platinum(II) is essential for the anticancer activity of platinum(IV) complexes (Hall and Hambley, 2002), so that adamplatin(II) (the analog of cisplatin) is presumably the complex that binds to DNA. The adducts of conventional cisplatin distort DNA conformation, inhibit replication and transcription (but they are also bypassed by DNA or RNA polymerases), and trigger apoptosis or necrosis (Fuertes et al., 2003). In addition, cisplatin adducts are removed from DNA mainly by NER. They are, however, also recognized by several proteins, such as HMG-domain proteins. The details of how the binding of HMG-domain proteins to cisplatin-modified DNA sensitizes tumor cells to cisplatin are still not completely resolved, but possibilities such as shielding cisplatin-DNA adducts from excision repair or these proteins being titrated away from their transcriptional regulatory function have been suggested as clues for how they are involved in antitumor activity (Brabec, 2002). The ability of HMGB1 protein, and probably other cisplatin-DNA-binding proteins, to influence the efficacy of the drug may be dependent on tumor cell type (Wei et al., 2003).

In broad terms, we have demonstrated that the DNA binding mode of adamplatin(II), including sequence preference, type of the major adducts, resulting conformational alterations, and their recognition by HMG-domain proteins, is not very different from that of cisplatin. In contrast, there are pronounced differences in several factors associated with the DNA binding mode, which may be associated with different biological effects of adamplatin(IV) and cisplatin.

The major adducts of the metabolite of adamplatin(IV) can inhibit DNA polymerization to a limited extent (although more efficiently than the same adducts of cisplatin), so that they can also be bypassed by DNA polymerases (Fig. 5). Likewise, the 1,2-GG intrastrand CLs of adamplatin(II) are removed from DNA by the components of mammalian NER system present in CFEs with lower efficiency than the CLs of cisplatin (Fig. 7), so that it is reasonable to expect that they may persist much longer on DNA than the adducts of cisplatin. Thus, the observation that the major adducts of adamplatin(II) impede elongation of DNA to a greater extent than those of cisplatin and that these adducts of adamplatin(II) are removed from DNA less easily may imply that these pharmacological factors are at least partially responsible for altered antitumor effects of this new platinum(IV) drug in comparison with conventional cisplatin.

The results of the present work also demonstrate that the monofunctional adducts of adamplatin(II) (in cell-free media) and adamplatin(IV) (in cells) readily cross-link proteins, apparently owing to the slow chelation reaction of monofunctional adducts of adamplatin(II). It is noteworthy that earlier observations have demonstrated that cisplatin also forms DNA-protein ternary CLs (Zhitkovich and Costa, 1992). However, the monofunctional adducts of cisplatin formed in the first step of the reaction with DNA close to bifunctional CLs with a relatively fast rate so that these adducts do not

![Fig. 8. DNA-protein cross-linking. Ternary complex formation of unmodified (lanes 1–4) and platinated oligodeoxyribonucleotide duplex (DP) containing single, site-specific 1,2-GG intrastrand CL of cisplatin (lanes 5–8) or adamplatin(II) (lanes 9–12); no protein added (lanes 1, 5, and 9), NF-xB added (lanes 2, 6, and 10), HMG1a added (lanes 3, 7, and 11), KF added (lanes 4, 8, and 12). Ternary complex formation was assessed by SDS-PAA gel electrophoresis. The nucleotide sequence of the duplex(DP) is shown in Fig. 1B.](image)

<table>
<thead>
<tr>
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<th>A2780cisR</th>
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<tr>
<td>Cisplatin</td>
<td>127 ± 7</td>
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<tr>
<td>Adamplatin(IV)</td>
<td>508 ± 10</td>
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TABLE 2
DNA-protein cross-linking in cells
DNA-protein CL formation following exposure of A2780 and A2780cisR cells for 48 h to cisplatin or adamplatin(IV). After washing, cells were subjected to the potassium-SDS procedure following overnight storage at −70°C. The cross-linking was evaluated using the DPC defined as a ratio of the amount of SDS-precipitable DNA in treated cells to the amount of SDS-precipitable DNA in control cells (Zhitkovich and Costa, 1992). The values of DPC shown are those normalized to the identical amount of platinum associated with the cells. Each value of DPC represents the average of four samples ± S.E. For other details, see the text.
persist for long enough to allow their extensive cross-linking to proteins to occur (Bancroft et al., 1990). Consistent with this conclusion is the relatively very low frequency of the DNA-protein ternary CLs produced in Chinese hamster ovary cells treated with cisplatin (<1%) (Plooy et al., 1984), although other studies revealed somewhat higher amounts of these ternary lesions in other types of tumor cells (Farrell et al., 2004). Adamplatin(II) forms more monofunctional adducts on DNA than cisplatin as a result of retardation of the rearrangement of its monofunctional to bifunctional adducts and favorable competitive reaction with protein. It is noteworthy that a decrease in this chelation rate was already reported for other cisplatin analogs bearing bulky hydrophobic nonleaving amine ligands, such as trans-1,2-diaminocyclohexanechloroplatinum(II) (Page et al., 1990) or cis-[N,N,N-2-amino-N-2-methylamino-2,2,1-bicycloheptane]dichloroplatinum(II) (Lambert et al., 1995), and is due to incorporation of steric hindrance by bulky hydrophobic amine ligand (Lambert et al., 1995). In aggregate, the capacity of DNA adducts of adamplatin(II) to cross-link proteins supports the idea that the capability of DNA adducts of this platinum compound to cross-link proteins represents an important feature of the mechanism underlying antitumor effects of this new platinum(IV) compound.

Other details of the mechanism of the DNA-protein-cross-linking by adamplatin(IV) or adamplatin(II) cross-linking are unknown. However, this cross-linking could result from a substitution of the labile ligand by a nucleophilic group of the protein such as a histidine, imidazole, or more likely a thioether group of a methionine or a thiol group of a cysteine, within the protein-DNA recognition complex.

The earlier reports have demonstrated that cross-linking proteins to platinum-DNA adducts markedly enhances the efficiency of these adducts to terminate DNA polymerization by DNA polymerases in vitro and to inhibit removal of these adducts from DNA by NER (Novakova et al., 2003). Hence, it is reasonable to suggest that DNA-protein ternary CLs produced by adamplatin could persist considerably longer than the non–cross-linked adducts, potentiating the toxicity of adamplatin(IV) toward tumor cells sensitive to this drug. Thus, covalent cross-linking of DNA and proteins by adamplatin represents a potential novel mechanism through which this compound could exert its antitumor activity.

The results of the present work also demonstrate that adamplatin(IV) reacts with GSH with the rate similar to that of parent cisplatin. In contrast, this new platinum(IV) compound is radically more potent also in the tumor cell line A2780cisR resistant to cisplatin (Kozubik et al., 2005), which is known to be resistant also through elevated levels of reduced GSH (Behrens et al., 1987). Hence, the unfavorable deactivation of adamplatin(IV) by GSH does not seem to be a strong determinant of cytotoxic effects of this new drug. It is noteworthy that MT-2 does not react with adamplatin(IV).

Complicating factors in the reaction of MTs with platinum(IV) drugs is their reduction by the cysteine thiols leading to cystine formation either intra- or intermolecularly and the many cysteine thiols in metallothioneins that are buried in metal-thiolate clusters (Zhong et al., 1997).

In conclusion, the results of the present work suggest that deactivation of adamplatin(IV) by sulfur-containing compounds (such as glutathione or metallothioneins) is likely to play a less significant role in the mechanism of resistance of tumor cells to adamplatin(IV) in contrast to the role of this factor in the effects of cisplatin. Moreover, the treatment of tumor cells with adamplatin(IV) does not result in DNA modifications that would be markedly different from those produced by cisplatin. Conversely, the effects of other pharmacological factors associated with antitumor effects of adamplatin(IV), such as enhanced accumulation of the drug in cells, strong inhibition of DNA polymerization by these adducts, enhanced persistence of the adducts due to their less efficient repair and DNA-protein cross-linking are different from the effects of these factors in the mechanism underlying activity of cisplatin. Hence, the differences between the effects of adamplatin(IV) and cisplatin observed in the present work may help understand the unique activity of adampatin(IV).

The cytotoxicity is the result of many events, beginning with the cell accumulation and proceeding through detoxification by thiols, DNA modification, and cellular responses to the DNA damage. It is not a simple task to reveal all aspects of the mechanism underlying antitumor effects of platinum complexes, and it is apparently incorrect to attribute cytotoxicity to their single property. Further studies are therefore warranted to reveal a relative contribution of all potential factors contributing to the potency of adamplatin(IV) in various types of cancer cells. Nevertheless, full clinical testing will only answer the fundamental question whether adamplatin(IV) represents a breakthrough in the development of antitumor platinum drugs.

Acknowledgments

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