Binding of Inositol 1,4,5-trisphosphate (IP₃) and Adenophostin A to the N-Terminal region of the IP₃ Receptor: Thermodynamic Analysis Using Fluorescence Polarization with a Novel IP₃ Receptor Ligand

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ABSTRACT

Inositol 1,4,5-trisphosphate (IP₃) receptors (IP₃R) are intracellular channels. Their opening is initiated by binding of IP₃ to the IP₃-binding core (IBC) and transmitted to the pore via the suppressor domain (SD). The major conformational changes leading to IP₃R activation occur within the N terminus (NT) and the IBC. We therefore developed a high-throughput fluorescence polarization (FP) assay using a newly synthesized analog of IP₃, fluorescein isothiocyanate (FITC)-IP₃, to examine the thermodynamics of IP₃ and adenophostin A binding to the NT and IBC. Using both single-channel recording and the FP assay, we demonstrate that FITC-IP₃ is a high-affinity partial agonist of the IP₃R. Conventional [3H]IP₃ and FP assays provide similar estimates of the Kₐ for both IP₃ and adenophostin A in cytosol-like medium at 4°C. They further establish that the isolated IBC retains the ability of full-length IP₃R to bind adenophostin A with ~10-fold greater affinity than IP₃. By examining the reversible effects of temperature on ligand binding, we established that favorable entropy changes (ΔS) account for the greater affinities of both ligands for the IBC relative to the NT and for the greater affinity of adenophostin A relative to IP₃. The two agonists differ more substantially in the relative contribution of ΔH and ΔS to binding to the IBC relative to the NT. This suggests that different initial binding events drive the IP₃R on convergent pathways toward a similar open state.

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Inositol 1,4,5-trisphosphate receptors (IP₃R) are intracellular channels that mediate release of Ca²⁺ from the endoplasmic reticulum by IP₃ (Foskett et al., 2007). All IP₃R are tetrameric and each subunit of approximately 2700 residues has an IP₃-binding core (IBC; residues 224–604 of IP₃R1) near its N terminus and six transmembrane domains toward the C terminus (Fig. 1A). The last pair of transmembrane domains with their intervening luminal loops from the four subunits form the pore. All three subtypes of vertebrate IP₃R and the single subtype in invertebrates share a similar structural organization. It remains unclear how IP₃ binding to the IBC opens the pore, but the N-terminal suppressor domain (SD; residues 1–223) is involved. Removal of the SD uncouples IP₃ binding from gating (Uchida et al., 2003), a region within the N terminus that includes the SD terminal suppressor domain (SD; residues 1–223) is involved. Removal of the SD uncouples IP₃ binding from gating (Uchida et al., 2003), a region within the N terminus that includes the SD interacts directly with residues close to the pore (Boehning and Joseph, 2000), and conformational changes initiated by IP₃ pass to the pore entirely via the SD (Rossi et al., 2009). These observations are presently supported by only limited knowledge of the structure of IP₃R. There are high-resolution structures of the SD (Bosanac et al., 2005) and of the IBC with IP₃ bound (Bosanac et al., 2002), and there are several, somewhat inconsistent low-resolution (~30 Å) structures of the entire IP₃R (Taylor et al., 2004). None of these structures of the entire IP₃R (Taylor et al., 2004). None of these structures of the entire IP₃R (Taylor et al., 2004). None of these structures of the entire IP₃R (Taylor et al., 2004). None of these
structures can yet provide specific insight into the conformational changes evoked by IP₃ binding, although small-angle X-ray scattering analyses are consistent with the idea that IP₃ causes the SD and IBC to adopt a more compact structure (Chan et al., 2007). Our recent results, derived from analysis of the energetics of agonist binding to IP₃R and its N-terminal fragments, suggest that major, as-yet-undefined conformational changes associated with activation of IP₃R by IP₃ occur within the N terminus (NT; residues 1–604) (Rossi et al., 2009).

Adenophostin A is a high-affinity agonist of IP₃R in which the essential bisphosphate moiety of IP₃ is retained but attached to a glucose rather than an inositol ring. The 2'-phosphate of adenophostin A mimics, at least in part, the 1-phosphate of IP₃ (Fig. 1B). Despite considerable effort, fuelled by synthesis of many adenophostin A-related analogs (Borissow et al., 2005; Mochizuki et al., 2006; Sureshan et al., 2008), the structural basis of the high-affinity binding of adenophostin A to IP₃R is unresolved. We have suggested that a cation-π interaction between the adenine moiety of adenophostin A and Arg-504 within the IBC may contribute to this high-affinity binding (Sureshan et al., 2009).

The thermodynamics of reversible ligand-receptor interactions can provide insight into ligand recognition and associated conformational changes (Borea et al., 2000) that may not be apparent in even high-resolution structures (Chaires, 2008; Olsson et al., 2008). For a spontaneous process, the entropy of the universe must increase (i.e., ∆G < 0) (Keeler and Wothers, 2006). This free energy change (∆G = ∆H − T∆S) comprises the entropy change of the system (∆S), arising from changes in the motions of water, ligand, and receptor and changes in the entropy of the surroundings (∆H/T) arising from changes in bonding (Williams et al., 2004). Because ∆G (and thereby Kᵢ) is often a balance between larger ∆H and T∆S, ligand-receptor interactions that differ minimally in ∆G may nevertheless be distinguished by measurements of ∆H and ∆S (Luque and Freire, 1998). But ∆S and ∆H are not independent because stronger bonds (large ∆H) more severely restrict the motions of the ligand, receptor, and perhaps water molecules (reduced ∆S); this is known as enthalpy-entropy compensation (Williams et al., 2004; Olsson et al., 2008). Despite the difficulty of trying to relate ∆H and ∆S directly to receptor-ligand structures (Chaires, 2008; Olsson et al., 2008), for at least some receptors these thermodynamic parameters distinguish agonist and antagonist binding (Weiland et al., 1979; Borea et al., 2000). See

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**Fig. 1.** FITC-IP₃ is a partial agonist. Key domains within a single subunit of IP₃R1, showing the NT, which comprises the IBC and SD and transmembrane domains (TMD) (A). The structure of the IBC with its α and β domains is shown with IP₃ bound (Protein Data Bank ID1N4K), highlighting the 2-O-atom (arrow, and enlarged alongside) to which fluorescein is attached by a short linker in FITC-IP₃. Structures of IP₃, FITC-IP₃, and adenophostin A (B). Effects of the indicated concentrations of FITC-IP₃ on specific binding of [³H]IP₃ (0.75 nM) to the NT (4 µg, 10 nM) and IBC (1.1 µg, 1 nM) in CLM (C and D). Results are means ± S.E.M., n = 3. Typical recordings from excised nuclear patches from DT40-IP₃R1 cells stimulated with 10 µM IP₃ or FITC-IP₃ in the pipette (E). C denotes the closed state, and the holding potential was 40 mV. Current (i)-voltage (V) relationship for single IP₃R stimulated with IP₃ or FITC-IP₃ (10 µM) (F). Single channel open probability (Pₒ) and mean channel open time (τₒ) for IP₃R stimulated with IP₃ or FITC-IP₃ (10 µM) (G). Results (F and G) are means ± S.E.M., n = 3–5.
Fluorescence polarization (FP) provides one means of examining the thermodynamics of ligand binding. When a rigid fluorophore is excited by plane-polarized light and it remains immobile during its fluorescence lifetime [4 ns for fluorescein 5-isothiocyanate (FITC)] (French et al., 1997), 60% of emitted light will be detected in the plane of the exciting light, and the anisotropy (A; see Materials and Methods) will be 0.4 (Serdyuk et al., 2007). But if the molecule rotates during its fluorescence lifetime, less emitted light will be aligned with the excitation plane, and A will be <0.4 (A = 0 if the fluorophore randomly reorients). Because the speed of tumbling is inversely proportional to molecular volume, binding of a small fluorescent probe (such as FITC-IP3, 0.85 kDa) to a much larger IP3R fragment (IBC or NT, 43–67 kDa) increases A. FP measures this change in A as a fluorescent ligand binds. It thereby allows nondestructive quantification of binding without the need to separate bound from free ligand: binding is measured without perturbing the equilibrium. This is useful for measuring of low-affinity interactions in which rapid ligand dissociation during separation of bound and free ligand compromises analysis. Additional advantages of FP include the opportunity to make many measurements from the same sample under different conditions (different temperatures in this analysis), cost-effectiveness, applicability to high-throughput analyses, and avoidance of radioisotopes. Here we synthesize FITC-IP3 and adenophootin A interactions with the NT and IBC.

Materials and Methods

Materials. Sources of most reagents were described previously (Rossi et al., 2009). Adenophootin A (Marwood et al., 2000) was synthesized and characterized as previously reported. IP3 was from Alexis Biochemicals (Nottingham, UK). The structures of the ligands used are shown in Fig. 1B. [3H]IP3 (681 GBq/mmol) was from PerkinElmer Life and Analytical Sciences (Beaconsfield, Buckinghamshire, UK). Pop-Culture was from Novagen (Beeston, UK). Glutathione Sepharose 4B beads, PD-10 columns, and GST-tagged PreScission protease were from GE Healthcare (Chalfont St. Giles, Buckinghamshire, UK).

Synthesis of FITC-IP3. We used FITC to label IP3, because FITC derivatives are available as isomerically pure 5- or 6-isomers at practical cost. To minimize the length of the linker for optimal FP, 2-O-(2-aminomethyl)-IP3 (Riley et al., 2004) was used for the conjugation with FITC. The short linker holds the reacting amine group close to the charged phosphate groups of IP3. This makes conjugation reactions of 2-O-(2-aminomethyl)-IP3 with active esters of dyes, such as succinimides, quite challenging under standard conditions because in aqueous buffer, the reaction is slow, leading to competing hydrolysis of activated dye, which must therefore be used in large excess. However, we found that in dry methanol with triethylamine as base, FITC reacted cleanly and selectively with 2-O-(2-aminomethyl)-IP3. The reaction was carried out in deuterated methanol in an NMR tube and monitored by 31P NMR spectroscopy.

To a solution of 2-O-(2-aminomethyl)-IP3 (Riley et al., 2004) (15 mg of triethylammonium salt, 20 μmol) in dry tetradeuteromethanol (0.75 ml) in an NMR tube was added dry triethylamine (50 μl) and a trace of EDTA (Na form), a 1H-decoupled 31P NMR spectrum taken at this stage showed three sharp signals at 5.0, 3.9, and 1.7 ppm. FITC (11.7 mg of solid, 30 μmol) was added, and the tube was sealed and shaken, then kept in the dark. A second 31P NMR spectrum taken after 4 h showed three new signals (5.1, 4.3, and 1.1 ppm) and indicated that the conjugation reaction was ~60% complete as judged by peak integrals. A third 31P NMR spectrum taken after 20 h showed that the reaction had progressed further (~70% complete). Additional FITC (8.3 mg, 20 μmol) was added. The next day (total time 48 h), the reaction was judged to be complete; only three signals remained in the 31P NMR spectrum. The contents of the NMR tube were washed into a round-bottomed flask with methanol and concentrated under reduced pressure. The residue was taken up in deuterated water (10 ml) and applied to a column of Q Sepharose Fast Flow resin (8 × 2 cm, bicarbonate form) that was protected from light. The column was washed well with deionized water, followed by aqueous triethylammonium bicarbonate (TEAB) until all unconjugated dye had eluted. This required a large volume of 0.6 M TEAB buffer (−1 L) until the eluent ran colorless. The column was then eluted using a linear gradient of TEAB (0.6–2.0 M over 250 ml) collecting 10-ml fractions. The target compound eluted between 0.9 and 1.2 M TEAB. The most intensely fluorescent fractions (five tubes) were combined and concentrated to give FITC-IP3 (triethylammonium salt) (Fig. 1B) as an orange glass. This was redissolved in deionized water, applied to a column of Chelex-100 resin (Na form, 2.5 ml), and eluted with deionized water. The product was lyophilized to give FITC-IP3 as the Na salt, which was accurately quantified by total phosphate assay (Ames and Dubin, 1960) (16 μmol, 80% yield); 31P NMR (Na form, D2O, 270 MHz): δ 5.82 (br s, 1H), 5.73 (br dd, δ = 8, 2 Hz, 1H), 5.73 to 5.77 (27, 1H, δ = 6.68 to 6.62 (m, 4H), 4.26 (ddd, appears as q, δ = 9.6, 9.4, 8.4 Hz, 1H), 4.12 (br s, 1H), 4.15 to 3.76 (m, 7H), 3.74 (dd, δ = 9.6, 2.7 Hz, 1H); 31P NMR (D2O, 109 MHz): δ 5.56 (1P), 5.18 (1P), 4.10 (1P); high-resolution mass spectometry (m/z) [M]+ calculated for C29H31N2O20P3S, 851.0321; found, 851.0330.

Expression and Purification of N-Terminal Fragments of IP3R1. The N-terminal fragments of rat IP3R1 (NT; residues 1–604; IBC, residues 224–604) were amplified by PCR from the full-length clone lacking the S1 splice site. PCR used primers P1 and P2 for the NT and P2 and P3 for the IBC. The sequences of all primers are listed in Supplemental Fig. 1A. The fragments are numbered by reference to the full-length (S1) rat IP3R1 (Genbank accession number GQ233032). Insertion of the S1 splice region into the IBC fragment used QuiChange mutagenesis kit (Stratagene, La Jolla, CA) with P4 and P5 primers (Supplemental Fig. 1A). The PCR products were ligated into the pGEX-6P-2 vectors (GE Healthcare) as BamHI/Xhol fragments to give pGEX-NT and pGEX-IBC. Both pGEX-NT and pGEX-IBC include an N-terminal GST linked to the IP3R fragment by a PreScission cleavage site. The sequences of all constructs were confirmed by DNA sequencing. The presence of the S1 splice site does not affect the Kd of the IBC for IP3 (Supplemental Fig. 1B). After cleavage from GST, the IP3R fragments retain only five non-native N-terminal residues (Fig. 2A). These are unlikely to affect IP3 binding because the Kd of the NT and IBC for IP3 are similar for these fragments and those prepared using thrombin cleavage of His6-tagged proteins (Rossi et al., 2009), in which only two non-native N-terminal residues (Gly-Ser) remain (data not shown). The constructs were transformed into Escherichia coli AVB101, 1 ml of the culture was grown for 12 h at 37°C in Luria-Bertani medium with 100 μg/ml ampicillin and then at 22°C until the OD600 reached 1 to 1.5. Protein expression was induced by addition of isopropyl-D-thiogalactoside at 15°C for 20 h. Cells were harvested (6000g, 5 min), and the pellet was suspended in Tris/EDTA medium (TE; 50 mM Tris and 1 mM EDTA, pH 8.3) supplemented with 10% Pop-Culture, 1 mM 2-mercaptoethanol, and protease inhibitor cocktail (Complete protease inhibitor, 1 tablet/50 ml; Roche Applied Science, Mannheim, Germany). The suspension was incubated with lysozyme (100 μg/ml) and RNAase (10 μg/ml) for 30 min on ice, and the lysate was then sonicated (Transonic T420 bath, 20 s; CamLab, Over, Cambridge). After centrifugation (30,000g, 60 min), 50 ml of supernatant was incubated with constant rotation for 30 min at 20°C with 0.5 ml of glutathione Sepharose 4B beads. The beads were transferred into an
empty PD-10 column and washed five times with 10 ml of Ca2+-free cytosol-like medium (CLM: 140 mM KCl, 20 mM NaCl, 2 mM MgCl₂, 1 mM EGTA, and 20 mM PIPES, pH 7.0) supplemented with 1 mM dithiothreitol at 4°C. The beads were then incubated with 0.5 ml (1 bed volume) of CLM supplemented with 1 mM dithiothreitol and 120 units/ml GST-tagged PreScission protease at 4°C for 12 h and the eluted IP₃-R fragment, free of PreScission, was collected. Protein concentrations were measured using the detergent-compatible assay with γ-globulin as standard (Bio-Rad Laboratories, Hemel Hempstead, Hertfordshire, UK). Cloning and expression of His₆-tagged IBC and NT fragments have been described previously (Rossi et al., 2000).

Protein samples were separated using NuPage precast 4–to-12% gels (Invitrogen). Samples were either silver-stained (Pierce Silver Stain Kit II; Thermo Fisher Scientific, Waltham, MA) or transferred to nitrocellulose membranes using the iBlot system (Invitrogen, Carlsbad, CA). The NT and IBC were identified using antisera raised to peptides corresponding to residues 62 to 75 (Cardy et al., 1997) or 326 to 343 (the first splice site) (Rossi et al., 2009) of IP₃-R, respectively.

**[^3H]IP₃ Binding.** Equilibrium-competition binding assays were performed at 4°C in a final volume of 500 μl of CLM containing[^3H]IP₃ (0.75 nM), purified protein (1–4 μM), and competing ligands. For some experiments, CLM was replaced by TEM. After 10 min, reactions were terminated by addition of 500 μl of ice-cold CLM containing 30% poly(ethylene glycol) 8000 and γ-globulin (600 μg) followed by centrifugation (20,000g, 5 min, 4°C). Pellets were solubilized in 200 μl of CLM containing 2% Triton X-100, mixed with EcoScanTA scintillation cocktail (National Diagnostics, Atlanta, GA) and the radioactivity was determined. Nonspecific binding determined by addition of 10 μM IP₃ or by extrapolation of competition binding curves to infinite IP₃ concentration gave indistinguishable results. Binding results were fitted to a Hill equation (Prism ver. 5; GraphPad Software, San Diego, CA) from which the IC₅₀, and thereby the Kᵦ and Bₘₐₓ, were calculated (Kenakin, 1997).

**Fluorescence Polarization.** FP measurements were performed in 96-well, half-area, black round-bottomed polystyrene microplates (Greiner Bio-One, Gloucester, UK) using a Pherastar plate reader (BMG Labtech, Aylesbury, UK) in a temperature-controlled chamber. An automated liquid-handling system (Qiagen; Qiagen, Crawley, West Sussex, UK) was used to prepare dilutions and to make most additions to plates. Periodic assessment established that the precision and reproducibility of the automated system (typically <5% error after eight serial dilutions) exceeded that of manual pipetting.

For saturation assays, serially diluted protein in CLM (0.3–400 nM IP₃-binding sites) was mixed with FITC-IP₃ (0.5 nM) in a final volume of 50 μl. For competition assays, serially diluted ligands in CLM were mixed with FITC-IP₃ (0.5 nM) and protein (80 nM for NT and 15 nM for IBC). Optimization of these conditions is described under Results. The plate was equilibrated at each temperature (4–37°C) for 20 min before measuring FP. There was negligible change in the pH of CLM across this temperature range (7.05 to 6.98), and because Ca²⁺ was omitted from the medium, there was no change in free [Ca²⁺]. Samples were excited by vertically polarized light at 485 nm, and emission was simultaneously measured at 538 nm in the horizontal and vertical planes. Each FP measurement was the average from 300 flashes delivered over ~5 s.

Fluorescence measurements were corrected for background fluorescence (<15% for the highest protein concentration, Supplemental Fig. 2) determined at each protein concentration in the absence of FITC-IP₃. Anisotropy (A) was calculated from these corrected fluorescence emission intensities in the vertical (Iᵥ) and horizontal (Iᵥ) planes (Jameson and Sawyer, 1995): A = (Iᵥ − IᵥIᵥ + 2Iᵥ).

Anisotropy of free FITC-IP₃ (Aᵥ) was determined in the absence of added protein, and anisotropy of bound FITC-IP₃ (AᵥB) was determined in the presence of saturating concentrations of IBC (100 nM) or NT (300 nM). The fraction of bound FITC-IP₃ (FᵥB) is related to the measured anisotropy (Aᵥ) by the following equation (Jameson and Sawyer, 1995): FᵥB = (AᵥB − Aᵥ)/AᵥB − Aᵥ.

Anisotropy caused by nonspecific binding (AᵥN) was determined by measuring anisotropy at each protein concentration in the presence of a saturating concentration of IP₃ (10 μM) (Aᵥ). Because the free FITC-IP₃ concentration is substantially reduced when it binds to the NT or IBC, Aᵥ overestimates the nonspecific binding.
in the absence of saturating IP₃. Our correction assumes that nonspecific binding is linearly related to the free FITC-IP₃ concentration: 

\[ A_{NS} = (A_1 - A_0)/(1 - F_H) \]

Anisotropy caused by specific binding (Aₛ) of FITC-IP₃ to IP₃R fragments was then calculated from Aₛ = A_M - A_NS, from which the fraction of specifically bound FITC-IP₃ (Fᴮₛ) was calculated and used for all subsequent analyses: 

\[ F_B = (A_S - A_0)/(A_B - A_0) \]

Calculation of Equilibrium Dissociation Constants from FP Analyses. To determine the Kᵯ of FITC-IP₃ for the IBC and NT, a fixed concentration of FITC-IP₃ (0.5 nM) was incubated with various concentrations of protein. The total protein concentration required to cause 50% of FITC-IP₃ (0.25 nM in our experiments) to be bound (R₅₀) was then determined. The free protein concentration required to bind 50% of FITC-IP₃ (Kᵯ) was then calculated by correcting for the amount of bound protein: Kᵯ = R₅₀ - 0.25 nM. The Kᵯ for IP₃ and adenophostin A were measured in equilibrium competition binding assays with FITC-IP₃. From the total concentration of competing ligand that caused a 50% decrease in specifically bound FITC-IP₃ (IC₅₀), the Kᵯ for competing ligands (Kᵯ) was determined at each temperature (Kenakin, 1997):

\[ Kᵯ = \frac{B \times I \times Kᵯ}{(L_T \times R_T) + B(R_T - L_T) + B - Kᵯ) \] (1)

where Kᵯ = Kᵯ for FITC-IP₃ at each temperature; \( L_T \) = total [FITC-IP₃]; \( R_T \) = total [NT or total [IBC]]; \( B = [NT/IBC - FITC-IP₃ \text{ complex}] \) at IC₅₀, calculated from B = \( L_T \times F_B \); and I = free [competing ligand] at IC₅₀, calculated from I = IC₅₀ - 0.5\( R_T \).

Calculation of \( \Delta S \) and \( \Delta H \) from \( Kᵯ \). The Kᵯ is related to changes in Gibbs free energy: \( \Delta G = R \cdot T \ln Kᵯ \), where R is the gas constant and T is absolute temperature. Assuming that the change in heat capacity (\( \Delta C \)) is temperature-independent, the enthalpy change (\( \Delta H \)), entropy change (\( \Delta S \)), \( \Delta C, \Delta G \), and T are related to the reference temperature (\( T₀ \), 296 K in our experiments) (Wittmann et al., 2009).

\[ \Delta G(T) = \Delta H(T₀)\times(T₀ - T) - T\times[\Delta S(T₀) + \Delta C(T₀)\times\ln\left(\frac{T}{T₀}\right)] \] (2)

Values for \( \Delta H, \Delta S, \) and \( \Delta C \) were determined by ordinary least-squares curve-fitting of \( \Delta G \) versus T (Prism software) (Motulsky and Christopoulous, 2003). Where \( \Delta H \) is unaffected by temperature (i.e., \( \Delta C = 0 \)) (Borea et al., 2000), the equation simplifies to the van’t Hoff equation (Wittmann et al., 2009), such that

\[ \ln Kᵯ = \frac{\Delta H}{R \cdot T} - \frac{\Delta S}{R} \] (3)

From which \( \Delta H/R \) can be determined from the slope of the plot of \( \ln Kᵯ \) versus \( 1/T \).

Single Channel Recording. Currents were recorded from patches excised from the outer nuclear envelope of DT40 cells expressing recombinant IP₃R1 using symmetrical cesium methanesulfonate (200 mM) as the charge carrier. The composition of recording solutions and methods of analysis were otherwise as described previously (Rahman et al., 2009).

Results

FITC-IP₃ Is a Partial Agonist of IP₃R. We used FITC-IP₃ (Fig. 1B) as the fluorescent ligand for FP assays. Fluorescein has an appropriate fluorescence lifetime (4 ns), synthesis of FITC-IP₃ is economical (see Materials and Methods), and FITC-IP₃ is relatively small (0.85 kDa) and so tumbles rapidly (it has a large rotational relaxation time).

Fluorescein was linked to IP₃ via its 2-O-position because the structure of the IBC with IP₃ bound has shown that the 2-hydroxyl of IP₃ is exposed and makes no significant contacts with the IBC (Bosanac et al., 2002) (Fig. 1A). This is consistent with structure-activity studies using full-length IP₃R and its N-terminal fragments in which 2-O-modified analogs of IP₃ retain biological activity (Potter and Lampe, 1995; Rossi et al., 2009). In equilibrium-competition binding assays in TEM to full-length IP₃R (KD for FITC-IP₃ = 3.25 ± 0.07 nM), the NT (1.34 ± 0.24 nM), or IBC (0.28 ± 0.06 nM), FITC-IP₃ completely displaced specific [³H]IP₃ binding with high affinity (Supplemental Fig. 3). In subsequent analyses, we focus on the NT (Ki, for FITC-IP₃ = 11.8 ± 0.2 nM) and IBC (2.0 ± 0.2 nM) in CLM (Fig. 1, C and D, and Table 1).
Our recent analysis of 2-O-modified IP₃ analogs established that they are partial agonists of IP₃R (Rossi et al., 2009). Single channel recordings from IP₃R1 expressed in the nuclear envelope of DT40 cells (Rahman et al., 2009) showed that FITC-IP₃ is also a partial agonist (Fig. 1E). FITC-IP₃ and IP₃ caused IP₃R to open to the same single-channel Cs⁺ conductance (γCs, ~220 pS; Fig. 1F), but the single-channel open probability (Pₒ) was lower with a maximally effective concentration of FITC-IP₃ (0.057 ± 0.01, n = 4) than with IP₃ (0.41 ± 0.04, n = 5) (Fig. 1G). The mean channel open time (τₒ, ~10 ms, Fig. 1, E and G) was the same for IP₃R stimulated with IP₃ and FITC-IP₃, indicating that IP₃R differ in the rates of channel opening when bound to the different ligands. IP₃R activated by the 2-O-modified partial agonists that we characterized previously also had τₒ similar to those activated by IP₃, but the partial agonists less effectively promoted channel opening (Rossi et al., 2009). We have suggested that the 2-O-substituents of these partial agonists disrupt transmission of an essential conformational change from the IBC to the SD and thereby reduce the amount of binding energy that is diverted into conformational changes of the protein. The lesser effect of removing the SD on FITC-IP₃ binding to the NT, relative to its effects on IP₃ and adenophostin A binding (Table 1), is consistent with FITC-IP₃ also disrupting communication between the IBC and SD. FITC-IP₃ is the weakest partial agonist of the IP₃R yet identified.

These results establish that IP₃ and FITC-IP₃ interact with the same binding site on the IP₃R and that FITC-IP₃ is a partial agonist. FITC-IP₃ is therefore a fluorescent ligand suitable for analysis of IP₃R behavior. In subsequent analysis, we use FITC-IP₃ to develop an FP assay to measure ligand binding to IP₃R fragments. We focus on the IBC and NT of IP₃R1 because the IBC is the minimal structure that binds IP₃ and its analogs (Bosanac et al., 2002), and major conformational changes associated with IP₃R activation occur within the NT (Rossi et al., 2009).

Optimization of a FP Assay for IP₃R. In radioligand binding assays, ligand depletion can be minimized by ensuring that the ligand concentration considerably exceeds that of the receptor. But the signal from a FP assay depends on a considerable fraction of the ligand (FITC-IP₃) being bound. This dictates that the concentration of ligand-binding sites is known accurately to allow both saturation binding analyses (where the fraction of bound FITC-IP₃ is determined as a function of protein concentration) and to compute free ligand concentrations in competition experiments. A second requirement for FP is therefore an accurately defined concentration of functional receptor.

After expression in bacteria, purification on glutathione Sepharose, and cleavage by PreScission (see Materials and Methods and Fig. 2A), the purified NT had the expected molecular mass (67 kDa) after SDS-polyacrylamide gel electrophoresis (Fig. 2B). Additional bands with lower molecular mass (~35 and ~50 kDa) were also detected by both silver staining and immunoblotting with an antisera to residues 62 to 75 (Fig. 2B). The smaller intensities of the three major bands were similar in immunoblots and after silver staining, indicating that the smaller proteins correspond to C-terminally truncated fragments of the NT. The largest of these (~51 kDa) probably terminates at approximately residue 456. Because the minimal IP₃-binding fragment terminates at residue 578 (Yoshikawa et al., 1996), only the complete NT (28 ± 7% of immunoreactive staining; equivalent to 37 ± 9% of total protein) is likely to bind IP₃. Similar analyses of purified IBC, where any truncation would abolish IP₃ binding (Yoshikawa et al., 1996), demonstrate that 24 ± 7% of the IP₃R1 immunoreactivity (32 ± 9% of total protein) has the size expected of the IBC (43 kDa) (Fig. 2C). Because heparin is a competitive antagonist of IP₃ at IP₃R, these interpretations are consistent with only the largest fragments being retained on heparin-agarose columns (Supplemental Fig. 4).

The concentration of functional NT was determined from [³H]IP₃ binding in both saturation binding (in TEM, Fig. 2D), where the reliability of the specific activity of [³H]IP₃ is critical, and in competition binding (in TEM and CLM, Fig. 2E), where the reliability of the specific activity of [³H]IP₃ is more critical. In parallel comparisons from a single preparation of NT (in TEM to maximize sensitivity), both assays provided similar estimates of the Kᵦ and Bₘₐₓ (Table 2). For both the IBC and NT, the Hill coefficient for IP₃ binding was ~1 (Table 2), consistent with noncooperative binding of IP₃ to a single class of site. These results confirm that equilibrium competition binding assays can be used reliably to determine the concentration of functional IP₃-binding sites in our preparations of the NT and IBC. In all subsequent analyses, concentrations of NT and IBC are derived from equilibrium competition binding with [³H]IP₃ in CLM (Table 2) and expressed as the concentration of IP₃-binding sites.

A high concentration of FITC-IP₃ is desirable to minimize background fluorescence (Checovich et al., 1995), but unless the concentration of FITC-IP₃ is kept well below its Kᵦ, there will be substantial depletion of free receptors in saturation binding experiments, and competition binding experiments will require high concentrations of precious protein and ligands. Supplemental Fig. 2 summarizes our optimization of protein and FITC-IP₃ concentrations for the FP assay. We use 0.5 nM FITC-IP₃ in all subsequent experiments, and the concentrations of IBC and NT described below.

### TABLE 1
Comparison of Kᵦ, determined by [³H]IP₃ binding and FP assays

<table>
<thead>
<tr>
<th>IBC</th>
<th>NT</th>
</tr>
</thead>
<tbody>
<tr>
<td>[³H]IP₃</td>
<td>FP</td>
</tr>
<tr>
<td>nM</td>
<td>nM</td>
</tr>
<tr>
<td>FITC-IP₃</td>
<td>2.0 ± 0.2</td>
</tr>
<tr>
<td>IP₃</td>
<td>8.7 ± 1.8</td>
</tr>
<tr>
<td>Adenophostin A</td>
<td>0.7 ± 0.1</td>
</tr>
</tbody>
</table>

### TABLE 2
IP₃ binding to the NT and IBC determined by saturation and competition binding assays

<table>
<thead>
<tr>
<th>TEM</th>
<th>CLM</th>
</tr>
</thead>
<tbody>
<tr>
<td>Saturation</td>
<td>Competition</td>
</tr>
<tr>
<td>NT</td>
<td>NT</td>
</tr>
<tr>
<td>Kᵦ (nM)</td>
<td>2.2 ± 0.5</td>
</tr>
<tr>
<td>Bₘₐₓ (nmol/mg)</td>
<td>1.5 ± 0.1</td>
</tr>
<tr>
<td>Hill coefficient</td>
<td>0.8 ± 0.1</td>
</tr>
</tbody>
</table>
Conventional and FP Analyses Provide Similar Estimates of $K_D$. The $K_D$ (12.5 ± 0.6 nM, $n = 3$) of the NT for FITC-IP$_3$ was determined by measuring the $A$ of FITC-IP$_3$ (0.5 nM) as a function of increasing concentration of NT in CLM at 4°C (Fig. 3A). This $K_D$ is similar to that obtained under identical conditions in competition with $[^3H]$IP$_3$ (11.8 ± 0.2 nM) (Fig. 3B).

From Fig. 3A, the lowest concentration of NT to give an almost maximal $A$ is 80 nM. Subsequent competition assays therefore used 0.5 nM FITC-IP$_3$ and 80 nM NT. Figure 3, C and D, shows FP competition assays with IP$_3$ and adenophostin A from which the $K_D$ for each was calculated (Table 1). These and similar results with the IBC (Supplemental Fig. 5) establish that when measured under identical conditions (CLM at 4°C), the FP and radioligand binding assays provide similar estimates of $K_D$ for the three key ligands (Table 1). The consistency persists across two different FP assays: saturation binding (for FITC-IP$_3$; Fig. 3A) and competition binding (for IP$_3$ and adenophostin A; Fig. 3, C and D). These results establish the utility of our FITC-IP$_3$-based FP assay for high-throughput analyses of ligand binding to IP$_{3R}$ fragments. In subsequent experiments, we exploit the uniquely nondestructive nature of the FP assay together with its ability to measure low-affinity interactions to examine the thermodynamics of ligand interactions with the IBC and NT in CLM.

**The IBC Retains High Affinity for Adenophostin A.** Adenophostin A has 10-fold lower $K_D$ than IP$_3$ for both the NT and IBC whether assessed by FP or in competition with $[^3H]$IP$_3$ (Table 1). The 10-fold difference is similar to that of the affinities of the two ligands for full-length IP$_3$R1 and to their relative potencies in evoking Ca$_{2+}$ release (Rossi et al., 2009). We reported previously that the NT and IBC differed more modestly in their relative affinities for IP$_3$ and adenophostin A (~3-fold) (Morris et al., 2002), and others obtained similar results with a shorter NT fragment (residues 1–581).

**Table 3**

<table>
<thead>
<tr>
<th>Temperature (K)</th>
<th>277</th>
<th>283</th>
<th>289</th>
<th>296</th>
<th>303</th>
<th>310</th>
</tr>
</thead>
<tbody>
<tr>
<td>IP$_3$ IBC</td>
<td>9 ± 0.6</td>
<td>12.6 ± 1.2</td>
<td>18.4 ± 2.2</td>
<td>23.5 ± 2.0</td>
<td>31.3 ± 2.2</td>
<td>52.9 ± 4.4</td>
</tr>
<tr>
<td>IP$_3$ NT</td>
<td>95 ± 7</td>
<td>119 ± 10</td>
<td>191 ± 23</td>
<td>281 ± 23</td>
<td>377 ± 38</td>
<td>490 ± 40</td>
</tr>
<tr>
<td>Adenophostin A IBC</td>
<td>0.9 ± 0.1</td>
<td>1.3 ± 0.2</td>
<td>1.8 ± 0.3</td>
<td>1.9 ± 0.2</td>
<td>2.2 ± 0.4</td>
<td>2.6 ± 0.4</td>
</tr>
<tr>
<td>Adenophostin A NT</td>
<td>8.6 ± 0.2</td>
<td>10.7 ± 0.2</td>
<td>16.6 ± 0.5</td>
<td>20.9 ± 0.3</td>
<td>27.9 ± 0.6</td>
<td>36.9 ± 3.2</td>
</tr>
</tbody>
</table>

**Fig. 4.** Thermodynamics of IP$_3$ and adenophostin A binding. FP competition binding assays with FITC-IP$_3$ (0.5 nM), IP$_3$, and either the NT (A, 80 nM) or IBC (B, 15 nM) at the indicated temperatures. Similar analyses for adenophostin A are shown in Supplemental Fig. 6. All results are summarized in Table 3. Effects of temperature on $\Delta G$ for IP$_3$ and adenophostin A binding to the NT and IBC (C). The lines are fitted using eq. 2 van’t Hoff plots for IP$_3$ and adenophostin A binding to the IBC and NT, where $K_D = 1/K_A$ (D). All binding analyses (A–D) were performed in CLM. Results (A–D) are means ± S.E.M. from three independent experiments, each with two replicates (many error bars are smaller than the symbols).
(Vanlingen et al., 2000). But both studies were performed in TEM-like media, and our study (Morris et al., 2002) used IP₃R fragments with an N-terminal His₆-tag, which we have shown to reduce substantially the affinity for IP₃ (Rossi et al., 2009). Our present results are significant because they demonstrate that both the NT (Glouchankova et al., 2000) and IBC retain the structural determinants of high-affinity adenophostin A binding. Furthermore, both IP₃ and adenophostin A bind with ~10-fold greater affinity to the IBC than to the NT (Table 1). This is consistent with our suggestion that full agonists of IP₃R, such as IP₃ and adenophostin A, divert substantial binding energy into rearrangement of the SD (Rossi et al., 2009). We conclude that interactions between the IBC and adenophostin A are sufficient to account for adenophostin A binding with ~10-fold greater affinity than IP₃ to IP₃R and that IP₃ and adenophostin A divert similar amounts of binding energy (~6 kJ/mol) into rearranging the relationship between the IBC and SD.

**Thermodynamics of Ligand Binding Analyzed by FP.** The Kᵣ for IP₃ in CLM at 4°C determined in competition with FITC-IP₃ is similar whether determined after 10 or 120 min from repetitive measurements of the same plate; the Kᵣ values (mean ± S.E.M.) for 10, 20, 30, 60, 90, and 120 min were 92 ± 23, 85 ± 16, 101 ± 22, 86 ± 11, 93 ± 19, and 97 ± 12 nM, respectively. This confirms the stability of the biological samples and establishes that equilibrium is attained within 10 min. A 20-min incubation was used for all subsequent analyses. To minimize variability in the thermodynamic analyses of ligand binding, we measured A from the same plate at different temperatures (4-37°C). The Kᵣ of IP₃ for the NT measured at 4°C (89 ± 7 nM, n = 3) was indistinguishable from that measured after first incubating the plate at 37°C for 20 min (Kᵣ = 490 ± 40 nM) and restoring it to 4°C (Kᵣ = 86 ± 9 nM). This shows that the effects of temperature on Kᵣ are fully reversible and justifies our use of the same plate for measurements at each temperature.

Figure 4 shows FP analyses for IP₃ binding to the NT and IBC at different temperatures. The results for the two full agonists, IP₃ and adenophostin A (Supplemental Fig. 6), are summarized in Table 3. With rare exceptions (Hannaert-Merah et al., 1994; Li et al., 2009), most analyses of ligand binding to IP₃R have been performed at 4°C and in media similar to TEM that have low ionic strength and/or high pH to maximize specific [³H]IP₃ binding. Our results provide the first quantitative analysis of ligand binding to IP₃R at different temperatures in cytosol-like media.

**TABLE 4** Thermodynamics of IP₃ and adenophostin A binding to the NT and IBC

<table>
<thead>
<tr>
<th></th>
<th>ΔG* (kJ/mol)</th>
<th>ΔH (kJ/mol · K)</th>
<th>ΔS (kJ/mol)</th>
<th>−TΔS* (kJ/mol)</th>
</tr>
</thead>
<tbody>
<tr>
<td>IP₃</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>NT</td>
<td>−37.1 ± 0.1</td>
<td>−37.0 ± 1.1</td>
<td>0.9 ± 3.7</td>
<td>−0.3 ± 1.1</td>
</tr>
<tr>
<td>IBC</td>
<td>−43.2 ± 0.1</td>
<td>−36.8 ± 0.3</td>
<td>21.3 ± 1.5</td>
<td>−6.3 ± 0.4</td>
</tr>
<tr>
<td>Adenophostin A</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>NT</td>
<td>−43.5 ± 0.02</td>
<td>−32.2 ± 1.1</td>
<td>35.6 ± 3.7</td>
<td>−10.5 ± 1.1</td>
</tr>
<tr>
<td>IBC</td>
<td>−49.4 ± 0.2</td>
<td>−21.9 ± 1.0</td>
<td>93 ± 3.6</td>
<td>−27.5 ± 1.1</td>
</tr>
</tbody>
</table>

* 296 K.

For three of the four interactions, the relationship between ΔG and temperature was approximately linear (ΔC ≈ 0). But there was some curvature in the relationship for binding of adenophostin A to the IBC (ΔC > 0) (Fig. 4C), perhaps suggesting increased exposure of a hydrophobic surface in the adenophostin A-IBC complex. Supplemental Table 1 provides our estimates of ΔC (derived from eq. 2), but the variances are large because calculating ΔC from the effect of temperature on Kᵣ introduces compound errors that compromise the analysis (Borea et al., 1998; Wittmann et al., 2009). We also analyzed the data using van’t Hoff plots (which assume that ΔC = 0; eq. 3) (Table 4, Fig. 4D). Estimates of ΔH and ΔS obtained by the two analyses (eqs. 2 and 3) were not significantly different (Table 4 and Supplemental Table 1). Both analyses indicate that changes in ΔS are the major component of the increased affinity of IP₃ and adenophostin A for the IBC relative to the NT (Fig. 5A) and the major component of the increased affinity of adenophostin A relative to IP₃ (Table 4, Fig. 5B).

**Discussion**

We have synthesized a fluorescent ligand of IP₃R (FITC-IP₃; Fig. 1B), demonstrated that it is a high-affinity weak partial agonist of IP₃R (Fig. 1), and used it to establish an FP assay that provides a high-throughput assay for analyses of ligand binding to IP₃R (Figs. 2 and 3). We have shown by both FP and conventional competition binding assays that the IBC alone is sufficient for high-affinity binding of adenophostin A (Table 1) and that for both IP₃ and adenophostin A, similar amounts of binding energy (~6 kJ/mol)
are diverted into rearranging the SD (Table 4 and Fig. 5A) (Rossi et al., 2009). Establishing the exact nature of this structural rearrangement will require high-resolution structures of the NT with and without ligand bound (see Introduction). FITC-IP₃, by contrast, diverts less binding energy (∼3.3 kJ/mol) into rearranging the SD. We suggest, by analogy with our extensive analyses of other 2-O-modified analogs of IP₃ (Rossi et al., 2009), that FITC-IP₃ is a partial agonist because the FITC moiety blocks effective communication between the IBC and SD, causing the channel to open less effectively.

The difference in affinities of IP₃ and adenophostin A for the IBC (ΔΔG) ∼6 kJ/mol (Fig. 5B) is comparable with the additional stability provided by a cation–π interaction (ΔG ∼2–10 kJ/mol) (Meyer et al., 2003) and is therefore consistent with our suggestion that only adenophostin A forms a cation–π interaction with the IBC. Binding of the two ligands to the IBC also differs in the relative contribution from ΔH and ΔTS. IP₃ binding is largely enthalpy-driven, whereas adenophostin A binding also involves a substantial entropy gain (Table 4, Fig. 5B). It is more difficult, without comparable studies of many additional analogs of IP₃ and adenophostin A, to account specifically for these different contributions of ΔH and ΔTS to ligand binding. The large enthalpy change for both ligands is likely to result from bonding between the phosphate groups of the ligands and charged residues in the IBC. We can consider several possible explanations for the substantial entropy gain associated with adenophostin A binding. Although some studies have correlated cation–π interactions with favorable ΔH (Sörme et al., 2005), ab initio analyses of cation–π interactions between adenine and an Arg residue suggest that the cation pair vibrates over larger distances than do the isolated partners (Biot et al., 2003). This increase in vibrational entropy might explain, at least in part, the increased entropy gain associated with adenophostin A binding. The larger ΔTS for adenophostin A relative to IP₃ binding would also be consistent with the larger apolar surface of adenophostin A, causing a greater entropy gain from ligand desolvation and hydrophobic interactions during binding (Teilum et al., 2009).

Adenophostin A binds to both the IBC and NT with 10-fold greater affinity than IP₃. The contributions of ΔH and ΔTS for binding of IP₃ and adenophostin A to the IBC are rather different. However, these contributions are more similar for the two ligands binding to the NT (Fig. 5B). This is consistent with the idea that both ligands are full agonists that bind differently but ultimately cause the channel to adopt an indistinguishable open state (Rossi et al., 2009).

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References


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