MONNA, a Potent and Selective Blocker for Transmembrane Protein with Unknown Function 16/Anoctamin-1

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ABSTRACT

Transmembrane protein with unknown function 16/anoctamin-1 (ANO1) is a protein widely expressed in mammalian tissues, and it has the properties of the classic calcium-activated chloride channel (CaCC). This protein has been implicated in numerous major physiological functions. However, the lack of effective and selective blockers has hindered a detailed study of the physiological functions of this channel. In this study, we have developed a potent and selective blocker for endogenous ANO1 in Xenopus laevis oocytes (xANO1) using a drug screening method we previously established (Oh et al., 2008). We have synthesized a number of anthranilic acid derivatives and have determined the correlation between biological activity and the nature and position of substituents in these derived compounds. A structure-activity relationship revealed novel chemical classes of xANO1 blockers. The derivatives contain a –NO₂ group on position 5 of a naphthyl group-substituted anthranilic acid, and they fully blocked xANO1 chloride currents with an IC₅₀ < 10 μM. The most potent blocker, N-((4-methoxy)-2-naphthyl)-5-nitroanthranilic acid (MONNA), had an IC₅₀ of 0.08 μM for xANO1. Selectivity tests revealed that other chloride channels such as bestrophin-1, chloride channel protein 2, and cystic fibrosis transmembrane conductance regulator were not appreciably blocked by 10–30 μM MONNA. The potent and selective blockers for ANO1 identified here should permit pharmacological dissection of ANO1/CaCC function and serve as potential candidates for drug therapy of related diseases such as hypertension, cystic fibrosis, bronchitis, asthma, and hyperalgesia.

Introduction

Calcium-activated chloride channels (CaCCs) have been found in a wide range of organisms and tissues. They have fundamental and wide-ranging physiological roles in functions such as epithelial secretion, sensory transduction and adaptation, nociception, regulation of smooth muscle cell contraction and vascular tone, and control of neuronal and cardiac excitability (Hartzell et al., 2005). Therefore, CaCCs are potential drug targets for diarrhea, asthma, cystic fibrosis, and hypertension (Verkman and Galietta, 2009). CaCCs were first described in the early 1980s, in Xenopus laevis oocytes, where they generate the fertilization potential that generates a fast electrical inhibition to prevent polyspermy (Miledi, 1982; Barish, 1983). However, the molecular identity of these channels remained elusive until the transmembrane protein with unknown function 16/anoctamin-1 (ANO1) was identified as a CaCC in 2005 (Caputo et al., 2008; Schroeder et al., 2008; Yang et al., 2008). Since then, ANO1 has rapidly garnered attention, and a number of reports have described the properties and physiological roles of this protein (Huang et al., 2009). Notably, a recent study revealed that ANO1 acts as a heat sensor to detect nociceptive thermal stimuli in sensory neurons and possibly mediate nociception (Cho et al., 2012).

The anoctamin family consists of 10 different protein subtypes. Among them, ANO1 has been the most extensively studied (Huang et al., 2009). ANO1 has very similar properties

ABBREVIATIONS: ANO1, anoctamin-1; CaCC, calcium-activated chloride channel; CFTR, cystic fibrosis transmembrane conductance regulator; CLC2, chloride channel protein 2; cRNA, complementary RNA; DCM, dichloromethane; DIDS, 4,4'-diisothiocyanatostilbene-2,2'-disulfonic acid; DMSO, dimethylsulfoxide; DPC, diphenylamine-2-carboxylic acid; hANO1, human ANO1; HEK293, human embryonic kidney 293; HR-ESI-MS, high-resolution electrospray ionization mass spectrometry; mBest1, mouse bestrophin-1; mCLC2, mouse chloride channel protein 2; MONNA, N-((4-methoxy)-2-naphthyl)-5-nitroanthranilic acid; NFA, niflumic acid; NPPB, 5-nitro-2-(3-phenylpropylamino)benzoic acid; 4TFMPA, N-((4-trifluoromethylphenyl)anthranilic acid; 4TFP4NA, N-((4-trifluorophenyl)-4-nitroanthranilic acid; xANO1, Xenopus laevis ANO1.
to endogenous CaCCs that have been observed in many different cells, tissues, and organisms. These properties include low-field-strength anion selectivity, Ca\(^{2+}\) sensitivity, voltage dependence, and pharmacological profile. Despite the physiological importance of ANO1, the lack of a potent and selective blocker for this protein has impeded a better understanding of the channel at the molecular, biophysical, and pharmacological level. Currently available blockers for CaCCs, including ANO1, include nifluamic acid (NFA), 4,4′-diiisothiocyantostilbene-2,2′-disulfonic acid (DIDS), 5-nitro-2-(3-phenylpropylamino)benzoic acid (NPPB), and melfloquine, all of which must be applied at high concentrations to completely block ANO1. The half-maximal concentrations for inhibition (IC\(_{50}\)) of NFA, DIDS, and NPPB are reported to be 37.3, 10.7, and 32.3 μM, respectively (Oh et al., 2008). Furthermore, these blockers are known to cause undesirable side effects and block other channels. For example, NFA and DIDS also block the volume-regulated anion channel in some cell types (Xu et al., 1997; Greenwood and Large, 1998), whereas all three, NFA, DIDS, and NPPB, have a blocking effect on the K\(^1\) channels. For example, NFA and DIDS also block the volume-regulated anion channel in some cell types (Xu et al., 1997; Greenwood and Leblanc, 2007). In addition, NFA, DIDS, and NPPB cause an elevation of intracellular Ca\(^{2+}\) concentration in several cell types, which can elicit other cellular responses (Reinsprecht et al., 1995; Shaw et al., 1995; Schultheiss et al., 2000). More recently, several ANO1 inhibitors, such as dichlorophen, benzbromarone, and hexachlorophene, have been identified with high-throughput screening methods, having IC\(_{50}\) values of 5.49, 9.97, and 10.0 μM, respectively (Huang et al., 2012). These compounds showed somewhat improved potency over conventional blockers but still fell short of submicromolar potency. In another high-throughput screening study, an aminophenylthiazole (T16A\(_{inh}\), ANO1; Namkung et al., 2011) was found to have an IC\(_{50}\) of around 1 μM, but no selectivity information was available. Thus, because of these issues related to low potency and selectivity, there is a very pressing need for improved ANO1 blockers.

Many attempts have been made to uncover chemical compounds that block the endogenous CaCC in X. laevis oocytes. In a previous study, we established an optimized protocol for large-scale drug screening using a two-electrode voltage-clamp recording system to search for better blockers for endogenous CaCCs in X. laevis oocytes (Oh et al., 2008), which were revealed to be dominantly mediated by endogenous ANO1 in X. laevis oocytes (xANO1) (Yang et al., 2008). In our previous study, we found a structural similarity between commercially available CaCC blockers and N-(4-trifluoromethylphenyl)anthranilic acid (4TFMPA), a novel potent blocker for xANO1, with an IC\(_{50}\) of 6.0 μM, which was synthesized based on structure-activity relationship analysis. In the present study, using the same screening method, we further examined the blocking effect of synthesized compounds using an in-depth structure-activity relationship analysis to discover new ANO1 blockers that have an IC\(_{50}\) less than 1 μM.

### Materials and Methods

#### Preparation of Oocytes

As described previously (Oh et al., 2008), mature stage V and VI oocytes were harvested from adult X. laevis females (Xenopus-I, Inc., Dexter, MI) that were maintained in an automated maintenance system (Xenopus System; Aquatic Habitats, Apopoka, FL). All experimental procedures described later were performed in accordance with the Korea Institute of Science and Technology (Seoul, Korea) institutional guidelines for humane animal handling. Animals were anesthetized by cooling with ice. Surgically removed ovarian follicles were treated with 2 mg/ml collagenase type IA at room temperature for 90 minutes in Ca\(^{2+}\)-free Barth’s solution containing 89 mM NaCl, 1.0 mM KCl, 2.4 mM NaHCO\(_3\), 0.82 mM MgSO\(_4\), and 10 mM HEPES (pH 7.4). Oocytes were extensively rinsed with normal Barth’s solution containing 88 mM NaCl, 1.0 mM KCl, 2.4 mM NaHCO\(_3\), 0.82 mM MgSO\(_4\), 0.53 mM Ca(NO\(_3\))\(_2\), and 5 mM HEPES (pH 7.4); placed in a culture of Barth’s solution containing 88 mM NaCl, 1.0 mM KCl, 2.4 mM NaHCO\(_3\), 0.82 mM MgSO\(_4\), 0.33 mM Ca(NO\(_3\))\(_2\), 0.91 mM CaCl\(_2\), 10 mM HEPES, 10 μg/ml streptomycin, and 10 μg/ml penicillin (pH 7.4); and maintained at 18°C. Oocytes were used ∼1–4 days after isolation.

#### Synthesis

Some anthranilic acid derivatives (compound 4f in Figs. 1 and 2F, and six compounds that are not designated in Fig. 2F) had been previously synthesized and reported (Oh et al., 2008). The other anthranilic acid derivatives were synthesized as described in Fig. 1.

**Step A.** II, IV, and VIII: Thionyl chloride was added dropwise to a solution of benzoic acid derivative (I, III, and VII) in anhydrous methyl alcohol at 0°C. After this addition, the mixture was stirred at reflux for 8–12 hours. The reaction mixture was basified with 10% sodium bicarbonate, and ethyl acetate was added. The organic layer was dried over anhydrous MgSO\(_4\), filtered, and the solvent was evaporated to give the products II, IV, and VIII.

**Step B.** IX: Triflic anhydride was added dropwise to a solution of 4-trifluoromethyl-2-hydroxybenzoic acid methyl ester (VIII) in pyridine and dichloromethane (DCM). This was stirred at room temperature for 6 hours and acidified by the addition of 1 M HCl\(_{aq}\). The mixture was extracted with DCM and washed with brine. The organic layer was dried over anhydrous MgSO\(_4\), filtered, and the solvent was evaporated. IX was purified by column chromatography. Amine derivatives (compound 2α–q, Supplemental Methods, compound synthesis 2) were prepared by this method, except for the commercially available amines.

**Step C.** V and X: Buchwald-Hartwig cross-coupling of amine (compound 2α–q, IV) and bromo (1b, II; or triflate (1a, IX)) gave anilinic acid methyl ester (V, X). Tris(dibenzylideneacetone) dipalladium (0)-chloroform adduct (0.05 eq) and (2-2′,6′-bis(diphenylphosphino)-1′,1′-binaphthalene (0.10 eq) in anhydrous toluene (3 ml) were stirred at room temperature for 30 minutes. Then, 2-bromo-N-nitrobenzoic acid methyl ester, Cs\(_2\)CO\(_3\) (1.4 eq), and aniline derivative (1.2 eq) were added and stirred at 110–130°C for 5–10 hours. The reaction mixture was filtered through Celite (Yakuri Pure Chemicals Co., LTD, Kyoto, Japan) and concentrated in vacuo. Anthranilic acid methyl ester (V, X) was purified by column chromatography.

**Step D.** VI: Iodination of bromo (V) gave VI. V, NaI (2 eq), CuI (0.05 eq), and trans-N,N-dimethylethanolamine-1,2-diamine (0.1 eq) in 1,4-dioxane were stirred at 110°C for 5 days. The reaction mixture was filtered through Celite and concentrated in vacuo. VI was purified by column chromatography.

**Step E.** 3, 4, 5, 6: Hydrolysis of benzoic acid methyl ester derivative (V, VI, and X) was accomplished by refluxing tetrahydrofuran/methanol (MeOH/CH\(_2\)\(_2\)OH (3:2) solution in the presence of lithium hydroxide. The reaction mixture was acidified by the addition of 1 M HCl\(_{aq}\) then extracted with ethyl acetate and washed with brine. The organic layer was dried over anhydrous MgSO\(_4\), filtered, and the solvent was evaporated to give anilinic acid derivatives (3, 4, 5, and 6; Supplemental Methods, compound synthesis 1).

#### Representative Examples of Active Compounds

N-(3-Methoxy-naphthyl)-5-Nitroanthranilic Acid (5p). The final step yields 94.6% (orange powder). Proton nuclear magnetic resonance spectroscopy (\(^{1}H\)-NMR, 400 MHz, dimethylsulfoxide (DMSO)–d\(_{6}\), 8.64 (dd, J = 2.7 Hz, 1H), 8.06 (dd, J = 9.4, 2.6 Hz, 1H), 8.06 (dd, J = 9.4, 2.6 Hz, 1H), 8.06 (dd, J = 9.4, 2.6 Hz, 1H), 8.06 (dd, J = 9.4, 2.6 Hz, 1H), 8.06 (dd, J = 9.4, 2.6 Hz, 1H).

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Fig. 1. General synthesis procedures, reagents, and conditions. Step A: trimethylsilyl chloride, MeOH, reflux; step B: OTf, pyridine, DCM, −78°C to room temperature; step C: 2,2′-bis(diphenylphosphino)-1,1′-naphthyl, Pd(dba)$_3$ · CHCl$_3$ (tris(dibenzylideneacetone)dipalladium-chloroform adduct), Cs$_2$CO$_3$, toluene, 110–130°C; step D: Cul, NaI, trans-N,N′-dimethylcyclohexane-1,2-diamine, 1,4-dioxane, 110°C; step E: LiOH, tetrahydrofuran/MeOH/H$_2$O (5:3:2), reflux.

1H), 7.84 (d, J = 8.7 Hz, 1H), 7.82 (d, J = 9.3 Hz, 1H), 7.53-7.49 (m, 1H), 7.39-7.35 (m, 1H), 7.19 (d, J = 1.9 Hz, 1H), 7.15 (d, J = 2.3 Hz, 1H), 6.82 (d, J = 9.5 Hz, 1H), 3.91 (s, 3H); $^{13}$C-NMR (100 MHz, DMSO-d$_6$) δ 169.4, 157.5, 154.0, 137.0, 135.8, 129.9, 128.9, 127.7, 125.0, 122.4, 115.7, 114.2, 111.4, 106.0, 55.9; high-resolution electrospray ionization mass spectrometry (HR-ESI-MS) m/z 337.0838 [M−H]$^-$.

N-(6-Methoxy)naphthyl)-5-Nitroanthranilic Acid (5r). The final step yields 90.1% (yellow powder). $^1$H-NMR (400 MHz, DMSO-d$_6$) δ 10.65 (bs, 1H), 8.75 (d, J = 2.8 Hz, 1H), 8.09 (dd, J = 9.4, 2.8 Hz, 1H), 7.84 (d, J = 8.0 Hz, 1H), 7.78 (d, J = 9.2 Hz, 1H), 7.56 (dd, J = 7.8, 7.6 Hz, 1H), 7.46 (s, 1H), 7.41 (d, J = 7.2 Hz, 1H), 7.22 (dd, J = 9.0, 2.4 Hz, 1H), 6.66 (d, J = 9.6 Hz, 1H), 3.90 (s, 3H); $^{13}$C-NMR (100 MHz, DMSO-d$_6$) δ 169.4, 158.2, 154.3, 143.5, 136.8, 136.4, 134.5, 129.8, 128.9, 127.2, 126.6, 124.9, 121.3, 120.0, 113.9, 111.2, 107.3, 55.8; HR-ESI-MS m/z 337.0823 [M−H]$^-$.

N-(4-Methoxy)-2-Naphthyl)-5-Nitroanthranilic Acid (5w). The final step yields 98.0% (orange powder). $^1$H-NMR (400 MHz, DMSO-d$_6$) δ 10.54 (bs, 1H), 8.74 (d, J = 2.6 Hz, 1H), 8.19 (dd, J = 9.4, 2.5 Hz 1H), 8.11 (d, J = 8.2 Hz, 1H), 7.84 (d, J = 8.0 Hz, 1H), 7.56–7.45 (m, 3H), 7.35 (d, J = 9.4 Hz, 1H), 6.95 (d, J = 9.5 Hz, 1H); $^{13}$C-NMR (100 MHz, DMSO-d$_6$) δ 169.2, 156.6, 152.7, 137.2, 136.7, 134.7, 129.8, 128.9, 127.8, 127.7, 125.5, 123.5, 122.0, 114.4, 113.2, 111.8, 102.8, 56.4; HR-ESI-MS m/z 337.0831 [M−H]$^-$.

N-((7-Methoxy)-2-Naphthyl)-5-Nitroanthranilic Acid (5z). The final step yields 94.9% (yellow powder). $^1$H-NMR (400 MHz, DMSO-d$_6$) δ 10.55 (bs, 1H), 8.73 (d, J = 2.8 Hz, 1H), 8.19 (dd, J = 9.4, 2.6 Hz, 1H), 7.91 (d, J = 8.6 Hz, 1H), 7.83 (d, J = 9.0 Hz, 1H), 7.78 (d, J = 1.3 Hz, 1H), 7.32–7.29 (m, 2H), 7.28 (d, J = 9.4 Hz, 1H), 7.14 (dd, J = 9.0, 2.4 Hz, 1H), 3.87 (s, 3H); $^{13}$C-NMR (100 MHz, DMSO-d$_6$) δ 169.2, 158.4, 152.7, 137.2, 136.8, 135.6, 129.8, 129.7, 128.9, 126.8, 121.1, 120.0, 118.7, 114.1, 111.7, 106.2, 55.7; HR-ESI-MS m/z 337.0830 [M−H]$^-$.

N-((8-Methoxy)-2-Naphthyl)-5-Nitroanthranilic Acid (5aa). The final step yields 91.1% (yellow powder). $^1$H-NMR (400 MHz, DMSO-d$_6$) δ 10.56 (bs, 1H), 8.74 (d, J = 2.6 Hz, 1H), 8.20 (dd, J = 9.4, 2.6 Hz, 1H), 8.03 (s, 1H), 7.98 (d, J = 8.6 Hz, 1H), 7.51 (d, J = 8.5 Hz, 1H), 7.43 (dd, J = 7.7, 7.7 Hz, 1H), 7.22 (d, J = 9.4 Hz, 1H), 7.01 (d, J = 7.5 Hz, 1H), 3.97 (s, 3H); $^{13}$C-NMR (100 MHz, DMSO-d$_6$) δ 169.1, 154.8, 152.8, 137.2, 135.8, 132.4, 129.9, 128.9, 126.5, 125.8, 124.1, 120.3, 115.3, 113.7, 111.7, 105.6, 56.1; HR-ESI-MS m/z 337.0833 [M−H]$^-$.

N-(4-Bromo)naphthyl)-5-Nitroanthranilic Acid (5ab). The final step yields 97.6% (brown powder). $^1$H-NMR (400 MHz, DMSO-d$_6$) δ 10.72 (bs, 1H), 8.77 (d, J = 2.7 Hz, 1H), 8.24 (d, J = 8.4 Hz, 1H), 8.09 (dd, J = 9.4, 2.8 Hz, 1H), 8.00–7.97 (m, 2H), 7.79 (dd, J = 7.7, 7.3 Hz, 1H), 7.70 (dd, J = 7.7, 7.2 Hz, 1H), 7.55 (d, J = 8.0 Hz, 1H), 6.72 (d, J = 9.4 Hz, 1H); $^{13}$C-NMR (100 MHz, DMSO-d$_6$) δ 169.3, 153.9, 137.2, 135.0, 132.6, 130.9, 130.6, 129.7, 129.0, 128.8, 128.5, 127.7, 124.4, 123.4, 120.5, 114.1, 111.7; HR-ESI-MS m/z 384.9819 [M−H]$^-$.
- Using the Warner model OC725B two-electrode voltage clamp amplifier
- Thapsigargin was then also removed by washing with the oocyte recording solution. All recordings were obtained at a holding potential of −60 mV. Synthesized drug derivatives were prepared in separate bottles and applied by gravity. The flow of solutions was approximately 1 ml/min.

### Electrophysiology of Cultured Astrocytes, Human Embryonic Kidney 293 Cells

Mouse astrocyte cell cultures were prepared as previously described (Park et al., 2009). Human embryonic kidney 293 (HEK293) cells were transfected with cDNA (human ANO1 (hANO1), mouse ANO1, or mouse chloride channel protein 2 (mCLC2)) expression plasmids tagged with enhanced green fluorescent protein using Effectene (Qiagen, Valencia, CA) or a blend of lipids (Fugene-6; Roche Molecular Biochemicals, Indianapolis, IN) at 1 μg of DNA per 35-mm culture dish. Single cells identified by enhanced green fluorescent protein fluorescence were used for whole-cell patch clamp experiments within 72 hours. Transfected HEK293 cells were recorded using a conventional whole-cell and an excised inside-out patch clamp. Fire-polished borosilicate glass patch pipettes were −3–5 MΩ. Experiments were conducted at room temperature (~20–25°C). To activate the channels directly, cultured astrocytes or HEK293 cells were patch clamped with a high Ca2+ intracellular solution that contained 146 mM CsCl, 5 mM CaCl2/EGTA/N-methyl-D-glucamine, 2 mM MgCl2, 8 mM HEPES, and 10 mM sodium succinate, at pH 7.3, adjusted with N-methyl-D-glucamine.

### Chemicals

Chemical compounds, including the following reagents, were purchased from Sigma-Aldrich (St. Louis, MO); collagenase type 1A, ionomycin-Ca2+ salt, thapsigargin, and cherythrine. HEPES was obtained from J. T. Baker (Mallinckrodt Baker, Inc., Phillipsburg, NJ).

### Data Analysis

Currents were digitally recorded with A xoScope software (Axon Instruments, Burlingame, CA) and Clampex 10.0 software. Data analysis was performed with SigmaPlot 10.0 (Systat Software, Inc., San Jose, CA). All current responses elicited during application of a blocker were normalized to the average of a Ca2+-induced Cl− current applied before application of the blocker. Normalized and averaged data were fitted using the SigmaPlot three-parameter logistic function to determine the dose-response relationship and the IC50. All data are expressed as the mean ± S.E.M. Statistical analyses were performed using a 2-tailed t test.

### Results

Pulled with a 9-7 programmable pipette puller (Sutter Instrument Co., Novato, CA). Microelectrodes had resistances of ~1–3 MΩ. During recordings, oocytes were continuously perfused with oocyte recording solution. All recordings were obtained at a holding potential of −60 mV. Synthesized drug derivatives were prepared in separate bottles and applied by gravity. The flow of solutions was approximately 1 ml/min.

#### Electricophysiologic of X. laevis Oocytes

To test the blocking effect of compounds on xANO1, xANO1 was activated by permeabilizing the plasma membrane to Ca2+ and changing the extracellular Ca2+ concentration. The oocyte plasma membrane was made permeable to Ca2+ by treatment with the Ca2+ ionophore ionomycin. Ca2+ sequestration by intracellular stores was inhibited with thapsigargin. Oocytes were incubated in an oocyte recording solution containing 96 mM NaCl, 2 mM KCl, 2 mM MgCl2, 0.5 mM EGTA, and 10 mM HEPES (pH 7.4), as well as 10 μM ionomycin, for a 30-minute period. After 30 minutes, the ionomycin was removed from the external solution by washing with oocyte recording solution. I onomycin-treated oocytes were subsequently incubated in the oocyte recording solution containing 1 μM thapsigargin for 90 minutes. Thapsigargin was then also removed by washing with the oocyte recording solution. Two-electrode voltage-clamp recordings were made using the Warner model OC725B two-electrode voltage clamp amplifier (Warner Instruments, Inc., Hamden, CT) with 1 M KCl-filled microelectrodes (1B150F-4; World Precision Instruments, Sarasota, FL).
Synthesized and purified compounds were tested for their blocking ability on xANO1 in Ca\(^{2+}\) ionophore–treated oocytes using the two-electrode voltage-clamp recording technique as described previously (Oh et al., 2008). We already showed that two repetitive calcium treatments induced currents with almost the same amplitude in our screening system (Oh et al., 2008). We confirmed that four repetitive calcium treatments induce stable currents with similar average amplitude and small variation (Fig. 2, A and B). We expanded our well-established recording method to permit simultaneous recording from four oocytes (Fig. 2C). A control ANO1 current in the oocyte was recorded by first adding 5 mM Ca\(^{2+}\) to the external bath for 5 seconds. Ca\(^{2+}\) was then removed and the drug derivatives added to the zero-Ca\(^{2+}\) solution before Ca\(^{2+}\) was added to activate ANO1. Current traces were recorded using the four-channel recording system. There was a small variation in peak current amplitudes. In response to repetitive 5-second applications of 5 mM Ca\(^{2+}\) plus treatment

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**Fig. 2.** Potency of synthesized derivatives with a phenyl group at the B position. (A) Representative trace of xANO1 currents induced by four repetitive 5 mM calcium treatments. (B) Summary of the experiments shown in A. Bar graph shows the stable currents with almost the same amplitude induced by four repetitive calcium treatments. Amplitude of the second, third, and fourth currents was normalized by the amplitude of the first current, respectively \((n = 7)\). Error bars indicate S.E.M. (C) Representative traces of xANO1 currents by simultaneous four-channel recording before and during application of 4TFP4NA. Oocytes were preincubated with 4TFP4NA for 30 seconds, and currents were induced by 5-second applications of extracellular Ca\(^{2+}\). The currents were measured at \(-60\) mV. (D) Representative dose-response relation of 4TFP4NA block of xANO1 currents \((n = 8)\). Error bars indicate S.E.M. (E) Structure of synthesized compound. (F) Summary table for substituents and the IC\(_{50}\) of tested compounds. Position A\(_2\) showed the most potent blocking effect. Comp. indicates compounds described in Fig. 1, which shows the synthesis procedures.
with the synthesized compound over the concentration range (~1–300 μM), peak amplitudes revealed a concentration-dependent block of the xANO1 current, and values for IC_{50} were obtained from dose-response curves. Representative dose-response relation of N-(4-trifluorophenyl)-4-nitroanthranilic acid (4TFP4NA) block of xANO1 currents is shown in Fig. 2D.

As we have previously shown that anthranilic acid derivatives with a para-substituted phenyl ring showed the highest blocking potency (Oh et al., 2008), we synthesized various anthranilic acid derivatives with para-positioned substituent groups on the phenyl ring (at position B in Fig. 2E) and with nitro (–NO₂) groups positioned at different locations on the benzoic acid ring (at positions A₁–A₃ in Fig. 2E). Figure 2F gives the IC_{50} values of test compounds. Derivative compounds having –NO₂ positioned at A₂ on the benzoic acid ring had a relatively high potency in all groups. Substituting different halogen groups on the phenyl ring (position B in Fig. 2E) revealed that potency increased with increasing size of the halogen (I > Br > Cl > F). For substituents with a trifluoromethyl group (–CF₃), –NO₂, or methoxy group (–OCH₃) on the phenyl ring, the most hydrophobic, that with –CF₃ was the most potent.

**Different Positional Effect Was Observed in Derivatives with Phenyl or Naphthyl Groups at the B Position.** As shown in Fig. 2F, the compounds with the greatest potency were produced when –NO₂ was located at position A₂ on the benzoic acid ring. We have already verified that 4TFMPA with a –CF₃ group para-positioned on the phenyl ring produced a better blocking effect compared with N-(4-nitrophenyl) anthranilic acid having a –NO₂ group located at the same position. Figure 2F shows that the derivative compound having –CF₃ on the phenyl ring had a higher potency than the derivative having –NO₂ on the phenyl ring. Therefore, we replaced the –NO₂ substituent group with –CF₃ at position A₂ on the benzoic acid ring (Fig. 3A). However, we found that derivative compounds with –NO₂ located at position A₃ on the benzoic acid ring were much more potent than compounds with –CF₃ at the same position (Fig. 3B).

Since compounds with more hydrophobic substituent groups had a higher potency in Fig. 2F, we further investigated additional derivative compounds possessing the more hydrophobic naphthyl group at position B (Fig. 3A) in combination with the benzoic acid ring. We synthesized a series of anthranilic acid derivatives having the N-a-naphthyl group. As expected, compounds with a naphthyl group were more potent than compounds with a phenyl group. In contrast to compounds containing the phenyl group, compounds with –NO₂ at position A₁ of the benzoic acid ring possessing the naphthyl group were demonstrated to be substantially potent blockers compared with compounds with –NO₂ at the A₂ position (Fig. 3C). These data support the idea that a close correlation exists between the potency of blocking xANO1 and the position of substituents on the benzoic acid ring or the hydrophobicity of substituents.

**Effect of Substituent Position in the Naphthyl Group on Blocking Potency.** As depicted in Fig. 3C, higher potency was observed for derivatives with 4-iodo, 4-bromo, and 4-methoxy substituents on the N-a-naphthyl group. To evaluate the positional effect, all positions on the naphthyl group (Fig. 4A) were substituted with an –OCH₃ group. Because it was easier...
to synthesize derivatives with \(-\text{OCH}_3\) than with iodine (\(-\text{I}\)), we prepared various compounds with \(-\text{OCH}_3\) at different positions. Figure 4B shows that all tested compounds conferred a relatively high potency that fully blocked the xANO1 chloride current with an IC\(_{50}\) < 10 \(\mu\)M. We further synthesized anthranilic acid derivatives with \(\beta\)-napthyl containing \(-\text{OCH}_3\) at different positions (Fig. 4C) and examined the blocking effect of these compounds for xANO1 (Fig. 4D). We also synthesized compounds that had bromine (\(-\text{Br}\)) and \(-\text{I}\) substituents at positions C\(_5\), C\(_{10}\), C\(_{13}\), or C\(_{14}\) (Fig. 4E). These positions were selected because they showed a relatively higher potency with \(-\text{OCH}_3\). We discovered that anthranilic acid derivatives with the \(\beta\)-napthyl group with \(-\text{OCH}_3\) at position C\(_{10}\), with \(-\text{I}\) at C\(_{10}\), \(-\text{Br}\) at C\(_{13}\), and \(-\text{I}\) at C\(_{13}\) had a profound blocking effect on xANO1, with IC\(_{50}\) values of 0.08, 0.90, 0.22, and 0.55 \(\mu\)M, respectively (Fig. 4E). The most potent blocker found, an anthranilic derivative having \(\beta\)-napthyl with an \(-\text{OCH}_3\) group at position C\(_{10}\), (N-(4-methoxy)-2-naphthyl)-5-nitroanthranilic acid) and with an IC\(_{50}\) of 0.08 \(\mu\)M, was named MONNA.

**Validation of Potent Blocker MONNA on the Steady-State Current.** In the blocker screening protocol, oocytes were treated with ionomycin to permeabilize the membrane to allow Ca\(^{2+}\) influx to activate xANO1. It is possible that the apparent blocking effect could be due to a direct block of ionomycin-mediated Ca\(^{2+}\) influx. To confirm that the prominent blocking effect of MONNA occurs at the ionophore or xANO1, we investigated the effect of MONNA on Ba\(^{2+}\)-induced currents carried by the ionophore (Fig. 5A). In response to application of Ba\(^{2+}\), a small current mediated by ionomycin was produced. This current was not blocked by MONNA or by 4TFMPA (Fig. 5B).

In our previous study, we showed that Cl\(^-\) currents elicited by Ca\(^{2+}\) influx in oocytes permeabilized with ionomycin consisted of a fast peak and a slow steady-state component (Oh et al., 2008). In addition, Schroeder et al. (2008) demonstrated that recording from ionomycin-treated *Axolotl* oocytes injected with xANO1 complementary RNA (cRNA) had calcium-activated currents with both fast and slow components, indicating that both components are mediated by xANO1. To examine whether tested blockers inhibit both the slow steady current as well as the fast current, we changed the test protocol. We measured current amplitudes from the baseline before the treatment of calcium to the steady-state current without MONNA and with MONNA over the concentration range of 1 nM to 100 \(\mu\)M in the presence of calcium (Fig. 5C). Concentration-dependent blocks of ANO1 currents were obtained by measuring the percentage of remaining current, and the IC\(_{50}\) was obtained from the dose-response curve (Fig. 5D). Although the IC\(_{50}\) for the slow current was slightly different from the IC\(_{50}\) for the fast current, MONNA still showed quite a high potency for blocking xANO1 (IC\(_{50}\)slow = 1.0 \(\mu\)M).

To examine the reversibility of MONNA, we tested the washout of MONNA. Calcium-induced peak current of xANO1 fully recovered after treatment with 0.1 \(\mu\)M MONNA (Fig. 5E). The percentage of current amplitude before MONNA treatment was 98.2 \(\pm\) 3.4% (Fig. 5F).

**MONNA Is a Selective Blocker for ANO1.** The inhibitory effect of MONNA on other ion channels was evaluated to establish an ion channel selectivity profile for this derivative compound. To verify that MONNA also blocks other members of the ANO1 protein family, inhibitory activity was assessed on hANO1 transiently expressed in HEK293 cells. hANO1 currents were measured by whole-cell patch-clamp recording in the presence of MONNA following 10 minutes of pretreatment with MONNA at a range of concentrations (0.1–10 \(\mu\)M; Fig. 6A). The IC\(_{50}\) for MONNA in hANO1 was 1.27 \(\mu\)M, as determined from the dose-response curve by measuring the current amplitude from +100 to –100 mV (Fig. 6B). There was no significant difference between the inward current and the outward current of hANO1 in terms of the block percentage (data not shown).

![Fig. 4. Effect of substituent position in the naphthyl group on blocking potency. (A and C) Structure of synthesized compound. (B, D, and E) Summary table for substituents and the IC\(_{50}\) of tested compounds. \(-\text{OCH}_3\) at position C\(_{10}\) of \(\beta\)-napthyl has the most potent blocking effect.]
We further examined the blocking effect of MONNA on other chloride channels such as bestrophin-1, CLC2, and cystic fibrosis transmembrane conductance regulator (CFTR). For mBest1, endogenously expressed in cultured mouse cortical astrocytes (Fig. 6D; Park et al., 2009, currents were recorded in the whole-cell patch-clamp mode with 4.5 mM Ca\(^{2+}\) in the patch pipette. MONNA (30 \(\mu\)M) was treated 10 minutes before and during patching. HEK293 cells transiently transfected with cDNA of human bestrophin-1 or mouse CLC2 (mCLC2) were recorded with the whole-cell patch-clamp technique. HEK293 cells were held at 0 mV and stimulated with voltage steps ranging from −100 to +100 mV. Current amplitudes were collected before and 2 minutes after the application of 10 \(\mu\)M MONNA. CFTR currents induced by isoproterenol were recorded with the two-electrode voltage-clamp method in X. laevis oocytes coinjected with cRNA encoding human CFTR and the β-adrenergic receptor. The oocytes were held at −60 mV, and 10 \(\mu\)M MONNA was added to the isoproterenol-containing solution (Fig. 6C). No significant blocking effect was observed with current responses in the presence of 30 \(\mu\)M MONNA for

Fig. 5. Validation of potent blocker MONNA on the steady-state current. (A) Representative traces to show blocking of Ba\(^{2+}\)-induced currents by MONNA and 4TFMPA in the ionophore. The current was measured at −60 mV. (B) Structure of tested compounds and summary bar graph of the experiments shown in A. Each number in the bar indicates the number of cells tested. Error bars indicate S.E.M. (C) Representative traces for test blocking effect of MONNA on the slow steady current component. A dashed line indicates the baseline. Current amplitudes were measured from baseline before the treatment of calcium to the steady-state current without MONNA and with MONNA over the concentration range in the presence of calcium. (D) IC\(_{50}\) and dose-response relation of MONNA for the slow current component of the Ca\(^{2+}\)-activated Cl\(^{−}\) current. Concentration-dependent block of ANO1 currents was obtained by measuring the percentage of remaining current, and IC\(_{50}\) was obtained from the dose-response curves. (E) Representative trace to show the washout of MONNA. A total of 0.1 \(\mu\)M MONNA was fully washed out after a 100-second wash with recording solution. (F) Summary bar graph of percentage current amplitude measured from the experiments shown in E (\(n = 9\)). Error bars indicate S.E.M. n.s., no significant statistical difference with the 2-tailed t test.
mBest1 and 10 μM MONNA for human bestrophin-1, mCLC2, and human CFTR (Fig. 6D).

Discussion

In this study, we have identified a novel class of ANO1 blocker that inhibits the endogenous ANO1 in X. laevis oocytes (xANO1), highlighting anthranilic acid derivatives with β-naphthyl possessing the substituent group –OCH₃, –I, or –Br. Notably, MONNA, an anthranilic derivative having β-naphthyl with an –OCH₃ group at C₁₀, is the most potent and selective blocker for xANO1 with an IC₅₀ of 0.08 μM.

Our drug screening method proved to be extremely efficient compared with the general blind screening for ion channel blockers. We validated the potent ANO1 blocker MONNA with less than 100 synthesized compounds. We were able to take advantage of target structure-focused libraries based on an anthranilic acid scaffold with specific substituents to arrive at the desired compounds because we began with a lead compound that was an ANO1 blocker. Therefore, fewer compounds required screening to obtain hit compounds. This strategy meant a smaller-scale, higher-quality screening process and dramatically reduced the subsequent hit-to-lead timescale.

Currently available blockers for ANO1 must be applied in high concentrations to completely block ANO1 and are not very selective. Many attempts have been made to discover blockers for CaCCs. In particular, recent reports have focused on the high-throughput screening of small molecules that inhibit the activity of ANO1 (Namkung et al., 2011; Huang et al., 2012). Although these recently screened small-molecule blockers have a high potency for ANO1 compared with conventional known blockers, the selectivity profile of these blockers should be determined.

The gating of ANO1 and the endogenous CaCC in oocytes is enigmatic and not completely understood (Kuruma and Hartzell, 2000; Xiao et al., 2011). However, all available evidence suggests that calcium-activated fast and slow currents are mediated by the same channel. Schröder et al. (2008) showed that recording from ionomycin-treated Axolotl oocytes injected with xANO1 cRNA had calcium-activated currents which had both fast and slow components. These two components with two distinct kinetics mediated by the xANO1 channel are probably associated with different conformational states; the fast current is the open state and the slow current may be a desensitized state. The conformational states of ion channels are believed to affect the affinity and efficacy of agonists and blockers. State-dependent affinity of ion channel blocker is a well-known phenomenon. Therefore, we can predict that different IC₅₀ values of MONNA to the fast and slow currents of xANO1 come from the differential conformational states.

The profile of this new compound MONNA, as a selective blocker of ANO1, is extraordinary. MONNA strongly blocked chloride currents in xANO1 as well as in hANO1, suggesting a common blocking effect of MONNA in other members of the ANO1 protein family. The level of sensitivity in endogenous xANO1 was different from that of hANO1 expressed in HEK cells. In our previous report, we confirmed that untransfected HEK293 cells showed negligible Ca²⁺-induced current response
and for possible therapeutic applications in those diseases beneficial for further understanding of the function of ANO1. The availability of a selective blocker for ANO1 should be significant for drug discovery and development. Meanwhile, the process should be significantly accelerated. From our current results, we are unable to distinguish whether the site of compound action on ANO1 is external or internal. However, a previous study revealed that diphenylamine-2-carboxylic acid (DPC) blocked from the outside and NFA blocked from both the inside and the outside when recordings were made in excised inside-out and outside-out patches from X. laevis oocytes (Qu and Hartzell, 2001). There is a possibility that blocking of NFA from the inside involves NFA crossing the membrane to block from the outside. Because the molecular structure of MONNA is similar to that of DPC and NFA, we can predict that MONNA might also block ANO1 in a manner that is similar to DPC or NFA. There is a possibility that MONNA can bind to the calcium binding site of ANO1 from the inside of the cells and influence the calcium dependence of ANO1. The crystal structure of ANO1 is not fully known, but when such structural information becomes available, the drug discovery process should be significantly accelerated. Meanwhile, the availability of a selective blocker for ANO1 should be beneficial for further understanding of the function of ANO1 and for possible therapeutic applications in those diseases associated with ANO1.

**Authorship Contributions**

**Participated in research design:** Oh, Hwang, Hartzell, Roh, Lee.  
**Conducted experiments:** Oh, Hwang, Jung, Yu, Kim, Choi.
MONNA, a potent and selective blocker for TMEM16A/Anoctamin-1.
Soo-Jin Oh, Seok Jin Hwang, Jonghoon Jung, Kuai Yu, Jeongyeon Kim, Jung Yoon Choi, H. Criss Hartzell, Eun Joo Roh, C. Justin Lee

Molecular pharmacology

Supplemental methods

Compound synthesis 1

N-(4-fluorophenyl)-3-nitroanthranilic acid (3a)
Final step yield 75.6 % (red solid). \(^1\)H – NMR (400 MHz, dimethyl sulfoxide (DMSO) – \(d_6\)) \(\delta\) 9.93 (bs, 1H), 8.19 (d, \(J = 7.5\) Hz, 1H), 8.05 (d, \(J = 7.8\) Hz, 1H), 7.08-7.06 (m, 3H), 6.98 (bs, 2H); \(^{13}\)C – NMR (100 MHz, DMSO – \(d_6\)) \(\delta\) 169.1, 161.5, 157.6, 140.0, 139.9, 138.4, 137.1, 131.3, 121.2, 121.1, 120.8, 119.1, 116.4, 116.1

N-(4-chlorophenyl)-3-nitroanthranilic acid (3b)
Final step yield 87.5 % (yellow solid). \(^1\)H – NMR (400 MHz, chloroform-D (CDCl\(_3\))) \(\delta\) 9.67 (bs, 1H), 8.25 (dd, \(J = 7.8\), 1.6 Hz, 1H), 8.10 (dd, \(J = 7.8\), 1.6 Hz, 1H), 7.26-7.22 (m, 2H), 7.00 (dd, \(J = 8.0\), 8.0 Hz, 1H), 6.95-6.91 (m, 2H); \(^{13}\)C – NMR (100 MHz, DMSO – \(d_6\)) \(\delta\) 168.7, 141.4, 140.9, 138.6, 137.1, 130.9, 129.5, 126.8, 122.4, 120.3, 120.1

N-(4-bromophenyl)-3-nitroanthranilic acid (3c)
Final step yield 87.2 % (yellow solid). \(^1\)H – NMR (400 MHz, DMSO – \(d_6\)) \(\delta\) 9.73 (bs, 1H), 8.18 (dd, \(J = 7.7\), 1.5 Hz, 1H), 8.10 (dd, \(J = 8.1\), 1.4 Hz, 1H), 7.38 (d, \(J = 8.7\) Hz, 2H), 7.16 (dd, \(J = 8.0\), 7.9 Hz, 1H), 6.87 (d, \(J = 8.7\) Hz, 2H); \(^{13}\)C – NMR (100 MHz, DMSO – \(d_6\)) \(\delta\) 168.7, 141.9, 141.1, 138.3, 137.0, 132.3, 130.9, 126.8, 120.5, 120.3, 114.6

N-(4-iodophenyl)-3-nitroanthranilic acid (3d)
Final step yield 94.7 % (yellow solid). \(^1\)H – NMR (400 MHz, DMSO – \(d_6\)) \(\delta\) 9.73 (bs, 1H), 8.18 (dd, \(J = 7.7\), 1.6 Hz, 1H), 8.09 (dd, \(J = 8.1\), 1.6 Hz, 1H), 7.52 (d, \(J = 8.7\) Hz, 2H), 7.16
**N-(4-trifluoromethylphenyl)-3-nitroanthranilic acid (3e)**

Final step yield 98.5 % (yellow solid) $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.56 (bs, 1H), 8.19-8.14 (m, 2H), 7.52 (d, $J = 8.0$ Hz, 1H), 7.33 (dd, $J = 7.8$, 7.7 Hz, 1H), 6.96 (d, $J = 7.9$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 168.1, 147.15, 143.3, 136.6, 136.1, 130.2, 129.1, 126.8, 126.4, 125.8, 123.7, 122.9, 122.1, 121.8, 121.5, 116.6, 60.2

**N-(4-nitrophenyl)-3-nitroanthranilic acid (3f)**

Final step yield 96.6 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.72 (bs, 1H), 8.15(dd, $J = 8.6$, 8.3 Hz, 2H), 8.06 (d, $J = 9.0$ Hz, 2H), 7.51 (dd, $J = 7.9$, 7.9 Hz, 1H), 6.82 (d, $J = 9.0$ Hz, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 167.2, 151.5, 146.0, 139.9, 136.1, 133.5, 129.7, 129.3, 126.2, 125.8, 114.8

**N-(4-methoxyphenyl)-3-nitroanthranilic acid (3g)**

Final step yield 88.7 % (orange solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.96 (bs, 1H), 8.18 (d, $J = 7.3$ Hz, 1H), 8.01 (d, $J = 7.5$ Hz, 1H), 7.00-6.96 (m, 1H), 6.90 (d, $J = 8.0$ Hz, 2H), 6.82 (d, $J = 7.8$ Hz, 2H), 3.71 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.3, 156.3, 140.9, 139.3, 138.2, 134.6, 131.5, 121.4, 119.4, 118.0, 114.8, 55.7

**N-(4-fluorophenyl)-4-nitroanthranilic acid (4a)**

Final step yield 85.1 % (red solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.70 (bs, 1H), 8.11 (d, $J = 8.7$ Hz, 1H), 7.65 (d, $J = 2.2$ Hz, 1H), 7.51 (dd, $J = 8.7$, 2.3 Hz, 1H), 7.42-7.38 (m, 2H), 7.32-7.27 (m, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.1, 161.0, 158.6, 151.3, 148.9, 136.0, 134.1, 126.2, 126.1, 117.3, 117.1, 116.9, 111.0, 107.7, 100.9, 99.1

**N-(4-chlorophenyl)-4-nitroanthranilic acid (4b)**

Final step yield 63.4 % (red solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.87 (bs, 1H), 8.12 (d, $J = 8.7$ Hz, 1H), 7.80 (d, $J = 1.8$ Hz, 1H), 7.54 (dd, $J = 8.7$, 2.0 Hz, 1H), 7.47 (d, $J = 8.6$ Hz, 1H), 7.17 (m, 2H), 7.01 (dd, $J = 8.7$, 8.7 Hz, 1H), 6.92 (d, $J = 8.7$ Hz, 2H), 6.82 (d, $J = 8.7$ Hz, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.1, 161.0, 158.6, 151.3, 148.9, 136.0, 134.1, 126.2, 126.1, 117.3, 117.1, 116.9, 111.0, 107.7, 100.9, 99.1
Hz, 2H), 7.37 (d, J = 8.6 Hz, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.0, 151.2, 147.7, 139.0, 134.1, 130.1, 128.6, 124.6, 118.5, 111.7, 108.4

$N$-(4-bromophenyl)-4-nitroanthranilic acid (4c)

Final step yield 89.6 % (red solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 9.80 (bs, 1H), 8.12 (d, J = 8.5 Hz, 1H), 7.82 (bs, 1H), 7.59 (d, J = 8.2 Hz, 2H), 7.55 (d, J = 8.4 Hz, 1H), 7.32 (d, J = 8.1 Hz, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.0, 151.2, 147.5, 139.4, 134.1, 133.0, 124.8, 118.5, 116.6, 111.8, 108.5

2-(4-iodophenylamino)-4-nitrobenzoic acid (4d)

Final step yield 97 % (brown solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 9.78 (bs, 1H), 8.12 (d, J = 8.6 Hz, 1H), 7.84 (s, 1H), 7.74 (d, J = 8.2 Hz, 2H), 7.55 (d, J = 8.5 Hz, 1H), 7.18 (d, J = 8.1 Hz, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.0, 151.2, 147.4, 139.9, 138.8, 134.0, 124.0, 118.6, 111.9, 108.5, 88.4

$N$-(4-trifluoromethylphenyl)-4-nitroanthranilic acid (4e)

Final step yield 91.7 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 9.90 (bs, 1H), 8.15 (d, J = 8.7 Hz, 1H), 8.04 (d, J = 2.2, 1H), 7.73 (d, J = 8.5 Hz, 2H), 7.66 (dd, J = 8.8, 2.2Hz, 1H), 7.51 (d, J = 8.4 Hz, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 168.7, 151.1, 145.9, 144.3, 134.0, 127.3, 126.2, 123.8, 123.5, 120.7, 120.3, 113.3, 110.1

$N$-(4-nitrophenyl)-4-nitroanthranilic acid (4f)

Final step yield 84.4 % (orange solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 10.04 (bs, 1H), 8.22-8.15 (m, 4H), 7.82 (dd, J = 8.7, 2.0 Hz, 1H), 7.42 (d, J = 9.0 Hz, 2H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 168.2, 150.9, 148.1, 143.8, 141.4, 133.9, 126.3, 123.7, 118.1, 115.6, 113.1

$N$-(4-methoxyphenyl)-4-nitroanthranilic acid (4g)

Final step yield 92.4 % (red solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 9.61 (bs, 1H), 8.09 (d, J = 8.7, 1H), 7.55 (d, J = 1.9, 1H), 7.43 (dd, J = 8.7, 1.8Hz, 1H), 7.27 (d, J = 8.7 Hz, 2H),
7.03 (d, $J = 8.7$ Hz, 2H), 3.80 (s, 3H) $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.3, 157.4, 151.3, 149.9, 134.0, 132.1, 126.4, 116.7, 115.5, 110.2, 107.2, 55.8

$N$-phenyl-4-nitroanthranilic acid ($4h$)

Final step yield 80.3 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 9.78 (bs, 1H), 8.12 (d, $J = 8.7$, Hz, 1H), 7.82 (d, $J = 2.3$ Hz, 1H), 7.51 (dd, $J = 8.7$, 2.3 Hz, 1H), 7.47-7.43 (m, 2H), 7.36-7.34 (m, 2H), 7.26-7.20 (m, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) 169.2, 151.3, 148.3, 139.7, 134.1, 130.3, 125.2, 123.2, 117.5, 111.1, 108.0

$N$-naphthyl-4-nitroanthranilic acid ($4i$)

Final step yield 93.7 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 10.20 (bs, 1H), 8.17 (d, $J = 8.7$, 1H), 8.05 (dd, $J = 7.9$, 1.8 Hz, 1H), 7.96 (d, $J = 7.4$ Hz, 1H), 7.91 (dd, $J = 8.1$, 1.8 Hz, 1H), 7.65-7.56 (m, 4H), 7.50 (dd, $J = 8.7$, 2.2 Hz, 1H), 7.44 (d, $J = 2.2$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.6, 151.3, 149.9, 135.4, 134.9, 134.1, 129.4, 129.1, 127.4, 127.2, 126.7(2), 122.3, 122.2, 117.1, 110.9, 107.9

$N$-((4-fluoro)naphthyl)-4-nitroanthranilic acid ($4j$)

Final step yield 88.5 % (brown solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 10.01 (bs, 1H), 8.15 (bs, 2H), 7.96 (d, $J = 5.4$ Hz, 1H), 7.70 (bs, 2H), 7.60 (bs, 1H), 7.47 (bs, 2H), 7.25 (bs, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.5, 157.8, 155.3, 151.4, 150.4, 134.0, 131.8, 131.1, 128.5, 127.9, 124.4, 124.3, 123.6, 123.5, 123.0, 121.2, 117.1, 110.8, 110.5, 110.3, 107.6

$N$-((4-chloro)naphthyl)-4-nitroanthranilic acid ($4k$)

Final step yield 98.1 % (red solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 10.20 (bs, 1H), 8.28 (d, $J = 8.3$, 1H), 8.18 (d, $J = 8.7$ Hz, 1H), 8.05 (d, $J = 8.4$ Hz, 1H), 7.81-7.77 (m, 2H), 7.73–7.69 (m, 1H), 7.63 (d, $J = 8.0$ Hz, 1H), 7.53 (dd, $J = 8.7$, 2.2 Hz, 1H), 7.45 (d, $J = 2.1$, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.5, 151.3, 149.4, 135.3, 134.1, 131.4, 130.5, 128.8, 128.3(2), 127.0, 125.0, 123.2, 122.1, 117.7, 111.4, 108.2
N-((4-bromo)naphthyl)-4-nitroanthranilic acid (4l)
Final step yield 94.5 % (red solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.23 (bs, 1H), 8.23 (d, $J = 8.4$, 1H), 8.18 (d, $J = 8.8$ Hz, 1H), 8.05 (d, $J = 8.5$ Hz, 1H), 7.97 (d, $J = 7.9$ Hz, 1H), 7.79 (dd, $J = 8.2$, 6.9 Hz, 1H), 7.70 (dd, $J = 7.6$, 7.5 Hz, 1H), 7.58 (d, $J = 8.0$ Hz, 1H), 7.53 (dd, $J = 8.7$, 1.8 Hz, 1H), 7.48 (d, $J = 1.8$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.5, 151.3, 149.2, 136.0, 134.1, 132.7, 130.7, 129.0, 128.3, 127.7, 123.2, 122.3, 119.1, 117.9, 111.4, 108.3

2-(4-iodonaphthalen-1-ylamino)-4-nitrobenzoic acid (4m)
Final step yield 87.2 % (orange solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.23 (bs, 1H), 8.20 (d, $J = 8.0$, Hz, 1H), 8.17 (d, $J = 8.6$ Hz, 1H), 8.09 (d, $J = 8.4$ Hz, 1H), 7.99 (d, $J = 8.3$ Hz, 1H), 7.74 (dd, $J = 7.4$, 7.0 Hz, 1H), 7.66 (dd, $J = 8.0$, 7.4 Hz, 1H), 7.58 (d, $J = 8.1$ Hz, 1H), 7.53 (s, 1H), 7.42 (d, $J = 7.7$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.5, 151.3, 149.1, 137.8, 136.8, 135.2, 134.0, 132.8, 130.1, 129.2, 128.2, 123.1, 122.6, 118.0, 111.5, 108.3, 96.3

N-((4-nitro)naphthyl)-4-nitroanthranilic acid (4n)
Final step yield 56.5 % (brown solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.85 (bs, 1H), 8.67 (d, $J = 8.6$ Hz, 1H), 8.42 (d, $J = 7.6$ Hz, 1H), 8.32 (d, $J = 8.3$ Hz, 1H), 8.22 (d, $J = 8.7$ Hz, 1H), 8.12 (s, 1H), 7.89 (dd, $J = 8.0$, 7.2 Hz, 1H), 7.80 (d, $J = 7.8$ Hz, 2H), 7.64 (d, $J = 8.5$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.0, 151.0, 145.5, 143.5, 140.5, 133.9, 130.7, 128.0, 127.1, 126.7, 123.9, 122.6, 121.9, 114.7, 112.4, 112.1

N-((4-methoxy)naphthyl)-4-nitroanthranilic acid (4o)
Final step yield 93.6 % (brown solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.88 (bs, 1H), 8.28-8.25 (m, 1H), 8.15 (d, $J = 8.7$ Hz, 1H), 7.85-7.83 (m, 1H), 7.61-7.58 (m, 2H), 7.53 (d, $J = 8.1$ Hz, 1H), 7.43 (dd, $J = 8.5$, 2.3 Hz, 1H), 7.14 (d, $J = 2.3$ Hz, 1H), 7.08 (d, $J = 8.2$ Hz, 1H), 4.04 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.5, 154.2, 151.4, 151.3, 134.0,
N-(4-fluorophenyl)-5-nitroanthranilic acid (5a)
Final step yield 88.8 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.38 (bs, 1H), 8.71 (d, $J = 2.8$ Hz, 1H), 8.16 (dd, $J = 9.4, 2.7$ Hz, 1H), 7.43-7.40 (m, 2H), 7.33- 7.29 (m, 2H), 6.92 (d, $J = 9.4$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.1, 161.6, 159.2, 153.3, 137.0, 135.0, 129.8, 128.9, 127.4, 127.3, 117.1, 116.9, 113.5, 111.3

N-(4-bromophenyl)-5-nitroanthranilic acid (5c)
Final step yield 91.2 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.45 (bs, 1H), 8.72 (d, $J = 2.7$ Hz, 1H), 8.17 (dd, $J = 9.4, 2.7$ Hz, 1H), 7.63 (d, $J = 8.6$ Hz, 2H), 7.33 (d, $J = 8.6$ Hz, 2H), 7.15 (d, $J = 9.4$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.0, 152.3, 138.2, 137.4, 133.0, 129.7, 128.8, 126.3, 118.1, 113.8, 112.0

2-(4-iodophenylamino)-5-nitrobenzoic acid (5d)
Final step yield 92 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.36 (bs, 1H), 8.71 (s, 1H), 8.18 (d, $J = 8.9$ Hz, 1H), 7.78 (d, $J = 8.3$ Hz, 1H), 7.20-7.16 (m, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.0, 152.2, 138.9, 139.4, 129.7, 128.8, 126.4, 114.0, 112.4, 90.3

N-(4-(trifluoromethyl)phenyl)-5-nitroanthranilic acid (5e)
Final step yield 95.4 % (orange solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 12.38 (bs, 1H), 8.80 (d, $J = 2.6$ Hz, 1H), 8.13 (dd, $J = 9.3, 2.7$ Hz, 1H), 7.72 (d, $J = 8.4$ Hz, 2H), 7.48 (d, $J = 8.3$ Hz, 2H), 7.38 (d, $J = 9.3$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.3, 150.9, 143.6, 138.0, 128.7, 128.1, 127.2, 126.1, 124.2, 123.4, 121.8, 118.6, 113.6

N-(4-nitrophenyl)-5-nitroanthranilic acid (5f)
Final step yield 73.5 % (red brown solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.92 (bs, 1H), 8.75 (bs, 1H), 8.29-8.25 (m, 3H), 7.59-7.55 (m, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.0, 152.3, 138.2, 137.4, 133.0, 129.7, 128.8, 126.3, 118.1, 113.8, 112.0
*N*-(4-methoxyphenyl)-5-nitroanthranilic acid (5g)

Final step yield 94.7 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.88 (bs, 1H), 8.97 (d, $J = 2.6$ Hz, 1H), 8.12 (dd, $J = 9.4$, 2.6 Hz, 1H), 7.21-7.18 (m, 2H), 6.99-6.97 (m, 2H), 6.88 (d, $J = 9.4$ Hz, 1H), 3.86 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.2, 158.0, 153.9, 136.5, 131.1, 129.7, 129.0, 127.0, 115.5, 113.3, 110.8, 55.8

*N*-phenyl-5-nitroanthranilic acid (5h)

Final step yield 90.5 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.38 (bs, 1H), 8.72 (d, $J = 2.8$ Hz, 1H), 8.18 (dd, $J = 9.5$, 2.8 Hz, 1H), 7.50-7.46 (m, 2H), 7.30-7.27 (m, 1H), 7.13 (d, $J = 9.5$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.1, 152.8, 138.6, 137.1, 130.3, 129.8, 128.9, 126.3, 124.5, 113.7, 111.5

*N*-naphthyl-5-nitroanthranilic acid (5i)

Final step yield 79.9 % (brown solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.72 (bs, 1H), 8.78 (d, $J = 2.7$ Hz, 1H), 8.10 (dd, $J = 9.4$, 2.6 Hz, 1H), 8.06 (d, $J = 7.3$ Hz, 1H), 7.96 (dd, $J = 6.4$, 2.4 Hz, 1H), 7.90 (d, $J = 7.6$ Hz, 1H), 7.65-7.57 (m, 4H), 6.69 (d, $J = 9.4$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.4, 154.3, 136.9, 134.8, 134.5, 129.8, 129.6, 129.1, 128.9, 127.6, 127.2, 126.6, 123.7, 122.5, 113.9, 111.4

*N*-(4-fluoronoaphthyl)-5-nitroanthranilic acid (5j)

Final step yield 86.7 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.51 (bs, 1H), 8.77 (s, 1H), 8.16 (d, $J = 8.0$, Hz, 1H), 8.07 (d, $J = 9.4$ Hz, 1H), 7.90 (d, $J = 7.8$ Hz, 1H), 7.75-7.67 (m, 2H), 7.61 (dd, $J = 7.0$, 5.2 Hz, 1H), 7.47 (dd, $J = 8.6$, 8.6 Hz, 1H), 6.56 (d, $J = 9.4$ Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.3, 158.3, 155.8, 154.6, 136.9, 131.1, 130.9, 129.7, 128.8, 128.7, 127.9, 124.9, 124.8, 124.3, 124.2, 123.1, 121.2, 113.9, 111.1, 110.5, 110.3

*N*-(4-chloronoaphthyl)-5-nitroanthranilic acid (5k)

Final step yield 61.7 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.77 (bs, 1H),
8.76 (d, J = 2.7 Hz, 1H), 8.28 (d, J = 8.4 Hz, 1H), 8.10 (dd, J = 9.4, 2.7 Hz, 1H), 7.96 (d, J = 8.2 Hz, 1H), 7.82-7.73 (m, 2H), 7.71 (dd, J = 7.8, 7.2 Hz, 1H), 7.61 (d, J = 8.0 Hz, 1H), 6.70 (d, J = 9.4 Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.4, 153.9, 137.2, 134.3, 131.4, 130.7, 129.7, 129.5, 128.8, 128.4, 126.9, 125.0, 123.9, 123.8, 123.3, 114.0, 111.6

$N$-((4-bromo)naphthyl)-5-nitroanthranilic acid (5l)

Final step yield 97.6 % (brown solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.72 (bs, 1H), 8.77 (d, J = 2.7 Hz, 1H), 8.24 (d, J = 8.4 Hz, 1H), 8.09 (dd, J = 9.4, 2.8 Hz, 1H), 8.00-7.97 (m, 2H), 7.79 (dd, J = 7.7, 7.3 Hz, 1H), 7.70 (dd, J = 7.7, 7.2 Hz, 1H), 7.55 (d, J = 8.0 Hz, 1H), 6.72 (d, J = 9.4 Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.3, 153.9, 137.2, 135.0, 132.6, 130.9, 130.6, 129.7, 129.0, 128.8, 128.5, 127.7, 124.4, 123.4, 120.5, 114.1, 111.7

$N$-((4-iodo)naphthyl)-5-nitroanthranilic acid (5m)

Final step yield 83.3 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.70 (bs, 1H), 8.76 (s, 1H), 8.22 (d, J = 7.7 Hz, 1H), 8.09 (dd, J = 8.0, 7.7 Hz, 2H), 7.91 (d, J = 8.2 Hz, 1H), 7.74 (dd, J = 7.6, 6.9 Hz, 1H), 7.65 (dd, J = 7.3, 7.2 Hz, 1H), 7.39 (d, J = 7.7 Hz, 1H), 6.72 (d, J = 9.3 Hz, 1H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.3, 153.9, 137.9, 137.2, 135.9, 135.2, 132.8, 130.5, 129.7, 129.3, 128.8, 128.5, 124.9, 123.4, 114.2, 111.8, 98.1

$N$-((4-methoxy)naphthyl)-5-nitroanthranilic acid (5n)

Final step yield 91.2 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.53 (bs, 1H), 8.76 (d, J = 2.8 Hz, 1H), 8.27-8.25 (m, 1H), 8.05 (dd, J = 9.4, 2.8 Hz, 1H), 7.80-7.77 (m, 1H), 7.60-7.58 (m, 2H), 7.51 (d, J = 8.1 Hz, 1H), 7.08 (d, J = 8.2 Hz, 1H), 6.48 (d, J = 9.5 Hz, 1H), 4.01 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.4, 155.3, 154.6, 136.5, 130.8, 129.7, 128.9, 128.0, 127.0, 126.6, 126.2, 125.4, 122.8, 113.7, 104.9, 101.0, 56.3

$N$-((2-methoxy)naphthyl)-5-nitroanthranilic acid (5o)

Final step yield 85.2 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.18 (bs, 1H),
8.75 (d, J = 2.8 Hz, 1H), 8.06-8.02 (m, 2H), 7.98 (d, J = 8.0 Hz, 1H), 7.68 (d, J = 8.4 Hz, 1H),
7.63 (d, J = 9.2 Hz, 1H), 7.51 (dd, J = 8.0, 6.8 Hz, 1H), 7.43 (dd, J = 7.6, 7.2 Hz, 1H), 6.21 (d,
J = 9.6 Hz, 1H), 3.90 (s, 3H); 
$^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.5, 154.7, 152.8, 136.6, 131.2, 129.6, 129.3, 128.9, 128.8, 127.9, 124.6, 121.9, 118.9, 114.7, 114.2, 110.7, 56.8

$N$-((5-methoxy)naphthyl)-5-nitroanthranilic acid (5q)

Final step yield 75.1 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.66 (bs, 1H),
8.77 (d, J = 2.8 Hz, 1H), 8.18 (d, J = 8.0 Hz, 1H), 8.10 (dd, J = 9.4, 2.8 Hz, 1H), 7.64-7.57
(m, 2H), 7.50 (dd, J = 8.4, 7.6 Hz, 1H), 7.44 (d, J = 8.5 Hz, 1H), 7.08 (d, J = 7.6 Hz, 1H),
6.67 (d, J = 9.4 Hz, 1H), 4.01 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 169.4, 155.8, 154.2, 136.9, 134.3, 130.7, 129.7, 128.9, 127.9, 126.6, 125.8, 124.2, 121.2, 114.4, 113.9, 111.1, 105.7, 56.2

$N$-((7-methoxy)naphthyl)-5-nitroanthranilic acid (5s)

Final step yield 86.1 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.67 (bs, 1H),
8.77 (d, J = 2.8 Hz, 1H), 8.12 (dd, J = 9.4, 2.8 Hz, 1H), 7.97 (d, J = 9.0 Hz, 1H), 7.87 (d, J =
8.1 Hz, 1H), 7.57 (d, J = 7.2 Hz, 1H), 7.45 (dd, J = 7.8, 7.7 Hz, 1H), 7.27 (dd, J = 9.0, 2.4 Hz,
1H), 7.18 (d, J = 2.3 Hz, 1H), 6.74 (d, J = 9.4 Hz, 1H), 3.78 (s, 3H); $^{13}$C - NMR (100 MHz,
DMSO – $d_6$) δ 169.5, 158.6, 154.0, 136.8, 133.4, 130.8, 130.7, 130.2, 129.7, 128.9, 127.2, 124.0, 123.8, 119.3, 113.8, 111.3, 101.1, 55.6

$N$-((8-methoxy)naphthyl)-5-nitroanthranilic acid (5t)

Final step yield 95.8 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 11.53 (bs, 1H),
8.75 (d, J = 2.8 Hz, 1H), 8.13 (dd, J = 9.5, 2.8 Hz, 1H), 7.74 (d, J = 7.6 Hz, 1H), 7.57-7.52
(m, 3H), 7.46 (dd, J = 8.0, 7.8 Hz, 1H), 7.29 (d, J = 9.5 Hz, 1H), 7.01 (d, J = 7.6 Hz, 1H),
3.87 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) δ 168.8, 156.2, 152.5, 137.1, 136.9, 134.6, 129.6, 129.1, 127.2, 126.6, 125.9, 121.6, 120.7, 119.3, 113.8, 112.4, 106.8, 56.0

$N$-((1-methoxy)-2-naphthyl)-5-nitroanthranilic acid (5u)
Final step yield 96.8 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ $\delta$ 10.62 (bs, 1H), 8.77 (d, $J = 2.8$ Hz, 1H), 8.21 (dd, $J = 9.4$, 2.8 Hz, 1H), 8.10 (d, $J = 8.2$ Hz, 1H), 7.98 (d, $J = 7.8$ Hz, 1H), 7.81 (d, $J = 8.7$ Hz, 1H), 7.63-7.59 (m, 2H), 7.57- 7.55 (m, 1H), 7.08 (d, $J = 9.4$ Hz, 1H), 3.83 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.3, 152.4, 148.7, 137.4, 132.8, 129.8, 128.8, 128.5, 127.4, 127.2, 126.5, 125.1, 123.9, 122.0, 114.0, 111.8, 61.7

$N$-((5-methoxy)-2-naphthyl)-5-nitroanthranilic acid (5x)

Final step yield 95.6 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 10.58 (bs, 1H), 8.74 (d, $J = 2.8$ Hz, 1H), 8.20 (dd, $J = 8.9$, 2.5 Hz, 1H), 7.86 (d, $J = 1.9$ Hz, 1H), 7.48-7.45 (m, 3H), 7.28 (d, $J = 9.4$ Hz, 1H), 6.96 (dd, $J = 4.4$, 4.3 Hz, 1H), 3.98 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.1, 155.3, 152.5, 137.2, 136.8, 135.2, 129.8, 128.8, 127.7, 123.9, 123.0, 122.7, 120.6, 120.0, 114.0, 111.8, 104.7, 56.1

$N$-((8-methoxy)-2-naphthyl)-5-nitroanthranilic acid (5aa)

Final step yield 91.1 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) 10.56 (bs, 1H), 8.74 (d, $J = 2.6$ Hz, 1H), 8.20 (dd, $J = 9.4$, 2.6 Hz, 1H), 8.03 (s, 1H), 7.98 (d, $J = 8.6$ Hz, 1H), 7.51 (d, $J = 8.5$ Hz, 1H), 7.43 (dd, $J = 7.7$, 7.7 Hz, 1H), 7.22 (d, $J = 9.4$ Hz, 1H), 7.01 (d, $J = 7.5$ Hz, 1H), 3.97 (s, 3H); $^{13}$C - NMR (100 MHz, DMSO – $d_6$) $\delta$ 169.1, 154.8, 152.8, 137.2, 135.8, 132.4, 129.9, 129.8, 128.9, 126.5, 125.8, 124.1, 120.3, 115.3, 113.7, 111.7, 105.6, 56.1

$N$-(4-chlorophenyl)-4-trifluoromethylanthranilic acid (6a)

Final step yield 95.6 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 9.74 (bs, 1H), 8.09 (d, $J = 8.2$ Hz, 1H), 7.45 (d, $J = 8.7$ Hz, 2H), 7.34-7.31 (m, 3H), 7.09 (d, $J = 8.3$ Hz, 1H)

$N$-(4-bromophenyl)-4-trifluoromethylanthranilic acid (6b)

Final step yield 97.2 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) $\delta$ 9.74 (bs, 1H), 8.09 (d, $J = 8.2$ Hz, 1H), 7.56 (d, $J = 8.7$ Hz, 2H), 7.33 (bs, 1H), 7.26 (d, $J = 8.7$ Hz, 2H), 7.10 (d, $J = 8.3$ Hz, 1H)

$N$-(4-nitrophenyl)-4-trifluoromethylanthranilic acid (6c)
Final step yield 99.3 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.97 (bs, 1H), 8.18 (d, $J = 9.1$ Hz, 2H), 8.13 (d, $J = 8.2$ Hz, 1H), 7.72 (bs, 1H), 7.40 (d, $J = 8.2$ Hz, 1H), 7.17 (d, $J = 9.2$ Hz, 2H)

$N$-(4-trifluoromethylphenyl)-4-trifluoromethylanthranilic acid (6d)

Final step yield 75.8 % (yellow solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 9.89 (bs, 1H), 8.12 (d, $J = 8.2$ Hz, 1H), 7.70 (d, $J = 8.4$ Hz, 2H), 7.58 (bs, 1H), 7.44 (d, $J = 8.4$ Hz, 2H), 7.23 (d, $J = 8.2$ Hz, 1H)

Compound synthesis 2

Synthesis of n-methoxy-1-naphthylamine

2-methoxy-1-nitronaphthalene (8). A mixture of 2-hydroxy-1-nitronaphthalene (500 mg, 2.64 mmol), dimethyl sulfate (0.5 ml, 5.3 mmol), and K$_2$CO$_3$ (73.1 mg, 5.3 mmol) in acetonitrile (10 ml) was refluxed for 1hrs. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was washed with water and brine and dried over anhydrous MgSO$_4$ and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (15:1) as the eluent to obtain the product (63.4 %, yellow solid). $^1$H – NMR (400 MHz, CDCl$_3$) δ 7.94 (d, $J = 9.2$ Hz, 1H), 7.82 (d, $J = 8.2$ Hz, 1H), 7.67 (d, $J = 8.6$ Hz, 1H), 7.61-7.57 (m, 1H), 7.46-7.42 (m, 1H), 7.32(d, $J = 9.2$ Hz, 1H), 4.02 (s, 3H)

2-methoxy-1-naphthylamine (2a). A mixture of 2-methoxy-1-nitronaphthalene (340 mg, 1.67 mmol) dissolved in MeOH (10 ml) and Raney Nickel (80 mg) was stirred at ambient
temperature under hydrogen balloon for 4 hrs. Product (90.5 %, brown solid) was filtered through Celite, and concentrated in vacuo. 1H – NMR (400 MHz, CDCl₃) δ 7.77-7.74 (m, 2H), 7.44-7.40 (m, 1H), 7.34-7.30 (m, 2H), 7.25 (d, J = 28 Hz, 1H), 4.23 (bs, 2H), 3.97 (s, 3H)

3-methoxy-1-naphthol (10). A dry hydrogen chloride was added for 10 min at 0 °C to a solution of 1,3-dihydroxyne(1 g, 6.24 mmol) in MeOH (5 ml). After 3.5 hrs at ambient temperature. The reaction mixture was basified with 1 N NaOH and ethyl acetate was added. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (4:1) as the eluent to obtain the product (69.9 %, orange solid) ¹H – NMR (400 MHz, CDCl₃) δ 8.07 (d, J = 8.4 Hz, 1H), 7.66 (d, J = 8.2 Hz, 1H), 7.42 (dd, J = 8.0, 7.0 Hz, 1H), 7.30 (ddd, J = 8.0, 7.2, 7.1 Hz, 1H), 6.73 (d, J = 1.6 Hz, 1H), 7.51 (d, J = 1.9 Hz, 1H), 6.23 (s, 3H), 3.91 (s, 3H)

(3-methoxy)naphthyltrifluoromethanesulfonate (2b). Triflic anhydride (0.78 ml, 4.58 mmol) was added dropwise in the course of 5 min to a mixture of 3-methoxy-1-naphthol (665 mg, 3.82 mmol) dissolved in DCM (10 ml) and pyridine (0.92 ml, 11.5 mmol) at 0 °C. After 3 hrs at ambient temperature. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (4 : 1) as the eluent to obtain the product (21.7 %, orange solid). ¹H – NMR (400 MHz, CDCl₃) δ 7.96 (d, J = 8.1 Hz, 1H), 7.76 (d, J = 8.3 Hz, 1H), 7.53 (dd, J = 7.9, 7.0 Hz, 1H), 7.46 (dd, J = 8.1, 7.1 Hz, 1H), 7.17-7.14 (m, 2H), 3.91 (s,
4-methoxy-1-naphthylamine (2c). A mixture of 4-methoxy-1-naphthylamine (600 mg, 2.95 mmol) dissolved in MeOH (10 ml) and 10 % Pd/C (100 mg) was stirred at ambient temperature under hydrogen balloon for 2 hrs. Product (94.9 %, violet solid) was filtered through Celite, and concentrated in vacuo. $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 8.06-8.02 (m, 2H), 7.43-7.41 (m, 2H), 6.74 (d, $J = 8.1$ Hz, 1H), 6.61 (d, $J = 8.1$ Hz, 1H), 5.17 (bs, 2H), 3.84 (s, 3H)

$N$-Boc-1-amino-5-hydroxynaphthalene (13). A mixture of 5-hydroxy-1-naphthylamine (1 g, 6.28 mmol), di–tert-butyl dicarbonate (2.06 g, 9.42 mmol) and guanidine hydrochloride (90 mg, 0.94 mmol) in EtOH (10 ml) was stirred for overnight. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated in vacuo to obtain the product. (92.8 %, wine solid). $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 10.13 (bs, 1H), 9.08 (s, 1H), 7.94 (d, $J = 8.4$ Hz, 1H), 7.51 (d, $J = 7.3$ Hz, 1H), 7.46 (d, $J = 8.5$ Hz, 1H), 7.37 (dd, $J = 8.2$, 7.6 Hz, 1H), 7.29 (dd, $J = 8.4$, 7.6 Hz, 1H), 6.86 (d, $J = 7.4$ Hz, 1H), 1.48 (s, 9H)

$N$-Boc-1-amino-5-methoxynaphthalene (14). A mixture of $N$-Boc-1-amino-5-
hydroxynaphthalene (1.13 g, 4.36 mmol), K$_2$CO$_3$ (1.21 g, 8.7 mmol), methyl iodide (0.54 ml, 8.72 mmol) in acetonitrile (10 ml) was reflux for 6 hrs. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO$_4$, filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with n-hexane/ethyl acetate (4:1) as the eluent to obtain the product (92.5 %, pink solid). $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 8.05 (d, $J = 8.4$ Hz, 1H), 7.92 (d, $J = 6.2$ Hz, 1H), 7.46-7.40 (m, 2H), 6.84 (ddd, $J = 5.4$, 2.5, 2.5 Hz, 1H), 6.81 (bs, 1H), 4.00 (s, 3H), 1.55 (s, 9H)

5-methoxy-1-naphthylamine (2d). TFA (trifluoroacetic acid, 5 ml) was added dropwise to a solution of $N$-Boc-1-amino-5-methoxynaphthalene (1.1 g, 4.02 mmol) in DCM (10 ml) at 0 °C. After 1 hrs at ambient temperature, the reaction mixture was basified with 1 N NaOH and DCM. The organic layer was dried over anhydrous MgSO$_4$, filtered and concentrated in vacuo to obtain the product (97.9 %, brown solid). $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 7.73 (d, $J = 8.4$ Hz, 1H), 7.39-7.34 (m, 2H), 7.27 (dd, $J = 7.6$, 7.4 Hz, 1H), 6.81-6.78 (m, 2H), 4.12 (bs, 2H), 3.97 (s, 3H)

$N$-Boc-1-amino-6-hydroxynaphthalene (16). Following the procedure of 13, the compound was prepared (90.7 %, yellow solid): $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 7.69 (d, $J = 9.0$ Hz, 1H), 7.59 (bs, 1H), 7.38-7.34 (m, 2H), 7.02-6.98 (m, 2H), 6.74 (bs, 1H), 6.04 (s, 1H), 1.57 (s, 9H)

$N$-Boc-1-amino-6-methoxynaphthalene (17). Following the procedure of 14, the compound was prepared (90.3 %, incarnadine solid): $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 7.78 (d, $J = 9.2$ Hz, 1H), 7.72 (d, $J = 6.4$ Hz, 1H), 7.51 (d, $J = 8.4$ Hz, 1H), 7.41 (dd, $J = 8.0$, 7.6 Hz, 1H),
7.18-7.14 (m, 2H), 6.78 (bs, 1H), 3.92 (s, 1H), 1.55 (s, 9H).

6-methoxy-1-naphthylamine (2f). Following the procedure of 2d, the compound was prepared (71.5 %, white solid): $^1$H – NMR (400 MHz, CDCl$_3$) δ 7.73 (d, $J = 10.0$ Hz, 1H), 7.27-7.20 (m, 2H), 7.13-7.10 (m, 2H), 6.65 (ddd, $J = 7.2, 1.2, 1.2$ Hz, 1H), 4.10 (bs, 2H), 3.92 (s, 3H) (89.9 %, brown solid): $^1$H – NMR (400 MHz, CDCl$_3$) δ 7.75 (bs, 1H), 8.96 (s, 1H), 7.75 (d, $J = 8.8$ Hz, 1H), 7.60 (d, $J = 8.0$ Hz, 1H), 7.36 (d, $J = 7.2$ Hz, 1H), 7.25-7.19 (m, 2H), 7.07 (ddd, $J = 8.8, 2.3, 2.3$ Hz, 1H), 1.48 (s, 9H) (80.1 %, brown solid): $^1$H – NMR (400 MHz, CDCl$_3$) δ 7.72 (d, $J = 8.8$ Hz, 1H), 7.28 (d, $J = 8.4$ Hz, 1H), 7.19-7.13 (m, 2H), 7.07 (d, $J = 2.8$ Hz, 1H), 6.79 (ddd, $J = 7.4, 1.2, 0.8$ Hz, 1H), 4.01 (bs, 2H), 3.94 (s, 3H)
8-hydroxy-1-naphthylamine (22). A saturated aqueous solution of NaHSO₃ (10 ml) was added to a solution of 1,8-diaminonaphthalene (1.5 g, 9.48 mmol) in EtOH (5 ml). The reaction mixture was refluxed for 24 hrs. Following addition of 6M aqueous solution of KOH (10 ml), the reaction mixture was heated to reflux for 2 hrs. The reaction mixture was acidified with 1 N HCl and ethyl acetate. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated in vacuo to obtain the product (86.1 %, brown solid). ¹H – NMR (400 MHz, CDCl₃) δ 7.48(dd, J = 8.3, 1.0 Hz, 1H), 7.30-7.28 (m, 2H), 7.25 (dd, J = 8.5, 7.0 Hz, 1H), 6.89 (dd, J = 7.3, 1.0, 1.0 Hz, 1H), 6.79 (dd, J = 4.4, 4.4 Hz, 1H)

N-Boc-1-amino-8-hydroxynaphthalene (23). Following the procedure of 13, the compound was prepared (81.1 %, orange solid): ¹H – NMR (400 MHz, CDCl₃) δ 9.99 (bs, 1H), 8.08 (bs, 1H), 7.44 (d, J = 8.1 Hz, 1H), 7.38 (dd, J = 7.5, 6.4 Hz, 2H), 7.26-7.22 (m, 2H), 6.80 (d, J = 7.0 Hz, 1H), 1.61 (s, 9H)

N-Boc-1-amino-8-methoxynaphthalene (24). Following the procedure of 14, the compound was prepared (92.8 %, orange solid): ¹H – NMR (400 MHz, CDCl₃) δ 9.87 (bs, 1H), 8.32 (ddd, J = 6.8, 2.1, 2.1 Hz, 1H), 7.42-7.39 (m, 3H), 7.31 (dd, J = 8.0, 7.9 Hz, 1H), 6.82 (d, J = 7.6 Hz, 1H), 4.03 (s, 3H), 1.55 (s, 9H)

8-methoxy-1-naphthylamine (2h). Following the procedure of 2d, the compound was prepared (84.1 %, brown solid): ¹H – NMR (400 MHz, CDCl₃) δ 7.31 (d, J = 8.2 Hz, 1H), 7.27-7.23 (m, 1H), 7.21 (dd, J = 7.9, 7.6 Hz, 1H), 7.10 (d, J = 8.1 Hz, 1H), 6.69 (d, J = 7.5 Hz, 1H), 6.57 (d, J = 7.5 Hz, 1H), 5.30 (d, J = 3.6 Hz, 1H), 3.97 (s, 3H)

Synthesis of n-methoxy-2-naphthylamine

\[
\begin{align*}
\text{O}_2\text{N} & \quad \text{HO} \\
\text{25} & \quad \text{CH}_3\text{CN} \quad \text{reflux} / 1 \text{ h} \\
\text{MeSO}_4 / \text{K}_2\text{CO}_3 & \quad \text{O}_2\text{N} \\
& \quad \text{OMe} \\
\text{26} & \quad \text{Reney nickel} / \text{H}_2 \text{ gas} \\
& \quad \text{MeOH} \quad \text{r.t.} / 4 \text{ h} \\
& \quad \text{H}_2\text{N} \quad \text{OMe} \\
\text{2i} & \quad \text{MeOH} \quad \text{r.t.} / 4 \text{ h}
\end{align*}
\]
2-nitro-1-methoxynaphthalene (26). Following the procedure of 8, the compound was prepared (78.1 %, brown solid): $^1$H – NMR (400 MHz, CDCl$_3$) δ 8.33-8.31 (m, 1H), 7.89 (d, $J = 8.7$ Hz, 2H), 7.69-7.65 (m, 3H), 4.15 (s, 3H)

1-methoxy-2-naphthylamine (2i). Following the procedure of 2a, the compound was prepared (94.1 %, brown solid): $^1$H – NMR (400 MHz, CDCl$_3$) δ 7.91 (d, $J = 8.4$ Hz, 1H), 7.70 (d, $J = 8.1$ Hz, 1H), 7.47-7.41 (m, 2H), 7.26-7.22 (m, 1H), 6.99 (d, $J = 8.7$ Hz, 1H), 3.95 (bs, 2H), 3.88 (s, 3H)

$\text{N-Boc-2-amino-3-hydroxynaphthalene (28).}$ Following the procedure of 13, the compound was prepared (75.5 %, brown solid): $^1$H – NMR (400 MHz, CDCl$_3$) δ 8.03 (s, 1H), 7.69-7.67 (m, 1H), 7.60-7.58 (m, 1H), 7.34-7.26 (m, 3H), 7.22 (s, 1H), 7.13 (bs, 1H), 1.56 (s, 9H)

$\text{N-Boc-2-amino-3-methoxynaphthalene (29).}$ Following the procedure of 14, the compound was prepared (94.9 %, yellow solid): $^1$H – NMR (400 MHz, CDCl$_3$) δ 8.42 (bs, 1H), 7.65 (dd, $J = 5.2$, 4.0 Hz, 1H), 7.56 (dd, $J = 7.5$, 5.1 Hz, 1H), 7.26-7.22 (m, 3H), 6.99 (s, 1H), 3.87 (s, 3H), 1.48 (s, 9H)

3-methoxy-2-naphthylamine (2j). Following the procedure of 2d, the compound was prepared (97.4 %, incarnadine solid): $^1$H – NMR (400 MHz, DMSO – $d_6$) δ 7.59 (d, $J = 7.5$ Hz, 1H), 7.46 (d, $J = 7.6$ Hz, 1H), 7.17-7.10 (m, 3H), 6.89 (s, 1H), 5.14 (bs, 2H), 3.89 (s, 3H)

$\text{N-Boc-2-amino-5-hydroxynaphthalene (31).}$ Following the procedure of 13, the compound
was prepared (95.5 %, brown solid): $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 8.09 (d, $J = 9.0$ Hz, 1H), 7.96 (bs, 1H), 7.34-7.29 (m, 2H), 7.24 (dd, $J = 8.2$, 7.9 Hz, 1H), 6.69 (d, $J = 7.4$ Hz, 1H), 6.65 (bs, 1H), 5.56 (s, 1H), 1.56 (s, 9H)

$N$-Boc-2-amino-5-methoxynaphthalene (32). Following the procedure of 14, the compound was prepared (42.7 %, brown liquid): $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 8.14 (d, $J = 9.0$ Hz, 1H), 7.96 (bs, 1H), 7.34-7.25 (m, 3H), 6.86 (bs, 1H), 6.68-6.66 (m, 1H), 3.94 (s, 3H), 1.54 (s, 9H)

5-methoxy-2-naphthylamine (2k). Following the procedure of 2d, the compound was prepared (87.7 %, brown liquid): $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 8.06 (d, $J = 8.7$ Hz, 1H), 7.26 (dd, $J = 8.1$, 7.8 Hz, 1H), 7.16 (d, $J = 8.2$ Hz, 1H), 6.91-6.87 (m, 2H), 6.56 (d, $J = 7.6$ Hz, 1H), 3.94 (s, 3H), 3.77 (bs, 2H)

$N$-Boc-2-amino-6-methoxynaphthalene (34). A mixture of 6-methoxy-2-naphthoic acid (1 g, 4.95 mmol), diphenylphosphoryl azide (1.22 ml, 6.18 mmol) and TEA (1.45 ml, 10.38 mmol) in tert-butanol (10 ml) was heated to reflux for overnight. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO$_4$, filtered and concentrated in vacuo to obtain the product (97 %, white solid). $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 7.91(bs, 1H), 7.65 (ddd, $J = 8.8, 2.1, 1.8$ Hz, 1H), 7.31 (ddd, $J = 8.8, 2.0, 2.0$ Hz, 1H), 7.11 (ddd, $J = 8.8, 2.5, 2.4$ Hz, 1H), 7.07 (d, $J = 2.3$ Hz, 1H), 6.56 (bs, 1H), 3.90 (s, 3H), 1.55 (s, 9H)

6-methoxy-2-naphthylamine (2l). Following the procedure of 2d, the compound was prepared (95.8 %, pink solid): $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 7.56 (d, $J = 8.6$ Hz, 1H), 7.50 (d, $J = 8.8$ Hz, 1H), 7.06 (ddd, $J = 8.9, 2.5, 2.5$ Hz, 1H), 7.03 (d, $J = 1.9$ Hz, 1H), 6.96 (d, $J =$
1.9 Hz, 1H), 6.93 (ddd, J = 8.5, 2.3, 2.3 Hz, 1H), 3.87 (s, 3H), 3.75 (bs, 2H)

**7-methoxy-2-naphthol (36).** Following the procedure of 8, the compound was prepared (48.3 %, white solid): \(^1\)H – NMR (400 MHz, CDCl\(_3\)) \(\delta\) 7.66 (d, J = 3.2 Hz, 1H), 7.64 (d, J = 3.6 Hz, 1H), 7.05 (d, J = 2.4 Hz, 1H), 6.99 (d, J = 2.5 Hz, 1H), 6.98 (bs, 1H), 6.94 (ddd, J = 8.7, 2.5, 2.5 Hz, 1H), 3.90 (s, 3H)

**7-methoxy-2-naphthylamine (2m).** A mixture of 7-methoxy-2-naphthol (600 mg, 3.44 mmol) and ammonium sulfite monohydrate (3.7 g, 27.52 mmol) in 28 % ammonium hydroxide (7 ml) was heated at 200 °C for 2 hrs. The reaction mixture was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO\(_4\), filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (3:1) as the eluent to obtain the product (73.5 %, white solid). \(^1\)H – NMR (400 MHz, CDCl\(_3\)) \(\delta\) 7.57 (d, J = 3.1 Hz, 1H), 7.56 (d, J = 3.1 Hz, 1H), 6.90-6.87 (m, 3H), 6.79 (ddd, J = 8.6, 2.1, 2.1 Hz, 1H), 3.88 (s, HCl 3H), 3.82 (bs, 2H)

**2-hydroxy-8-iodonaphthalene (38).** A cold solution of NaNO\(_2\) (260 mg, 3.77 mmol) in
water (1 ml) was slowly added to a solution of 2-hydroxy-1-naphthylamine (500 mg, 3.14 mmol) in conc. HCl (0.3 ml) and water (1 ml) at 0 °C and the reaction mixture was stirred 2 hrs at 0 °C. When diazotization was complete, a hot solution of KI (935.5 mg, 5.65 mmol) in water (1.5 ml) at 60 °C was added slowly and suspension was heated at 70 °C for 1 hr. The reaction mixture was basified with 1 N NaOH and ethyl acetate was added. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (4:1) as the eluent to obtain the product (47.7 %, yellow solid). ¹H – NMR (400 MHz, CDCl₃) δ 8.02 (d, J = 7.3 Hz, 1H), 7.75 (d, J = 8.1 Hz, 1H), 7.69 (d, J = 8.8 Hz, 1H), 7.44 (d, J = 2.3 Hz, 1H), 7.12 (ddd, J = 8.8, 2.4, 2.4 Hz, 1H), 7.04 (dd, J = 8.0, 7.5 Hz, 1H), 5.15 (s, 1H)

2-amino-8-iodonaphthalene. (39). Following the procedure of 2m, the compound was prepared (82.3 %, white solid): ¹H – NMR (400 MHz, CDCl₃) δ 7.94 (d, J = 7.3 Hz, 1H), 7.64 (d, J = 8.0 Hz, 1H), 7.54 (d, J = 8.6 Hz, 1H), 7.21 (d, J = 2.1 Hz, 1H), 6.92-6.88 (m, 2H), 3.92 (bs, 2H)

8-methoxy-2-naphthylamine (2n). A mixture of 2-amino-8-iodonaphthalene (660 mg, 2.45 mmol), CuI (46.7 mg, 0.25 mmol), Cs₂CO₃ (1.12 g, 3.43 mmol) and 1,10-phenanthroline (88.3 mg, 0.49 mmol) in MeOH (8 ml) was heated at 110 °C for 3 d. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (4:1) as the eluent to obtain the product (60.8 %, brown liquid). ¹H – NMR (400 MHz, CDCl₃) δ 7.61 (d, J = 8.6 Hz, 1H), 7.39 (d, J = 2.3 Hz, 1H), 7.29 (d, J = 8.2 Hz, 1H), 7.13 (dd, J = 8.0, 7.8 Hz, 1H), 6.94 (ddd, J = 8.6, 2.4, 2.4 Hz, 1H), 6.74 (d, J = 7.6 Hz, 1H), 3.97 (s, 3H), 3.84 (bs, 2H)
Synthesis of m-halogen-n-naphthylamine

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\begin{align*}
\text{O} & \quad \text{OH} \\
\text{DPPA/TEA} & \quad \text{reflux/overnight} \\
\text{rBuOH} & \\
\text{40} & \rightarrow \\
\text{NHBoc} & \\
\text{F} & \\
\text{41} & \rightarrow \\
\text{NH}_{2} & \\
\text{F} & \\
\text{2o} &
\end{align*}
\]

*N-Boc-1-amino-4-fluoronaphthalene. (41).* Following the procedure of 34, the compound was prepared (83.9 %, white solid): \( ^1 \text{H} \) – NMR (400 MHz, CDCl\(_3\)) \( \delta \) 8.13-8.11 (m, 1H), 7.90-7.88 (m, 1H), 7.71 (bs, 1H), 7.61-7.54 (m, 2H), 7.12 (ddd, \( J = 10.0, 8.4, 8.4 \) Hz, 1H), 6.65 (bs, 1H), 1.55 (s, 9H)

4-fluoro-1-naphthylamine (2o). Following the procedure of 21, the compound was prepared (98 %, brown solid): \( ^1 \text{H} \) – NMR (400 MHz, CDCl\(_3\)) \( \delta \) 8.07-8.05 (m, 1H), 7.81-7.79 (m, 1H), 7.53-7.49 (m, 2H), 6.96 (ddd, \( J = 10.2, 8.3, 8.2 \) Hz, 1H), 6.64 (ddd, \( J = 8.1, 4.4, 4.4 \) Hz, 1H), 3.94 (bs, 2H)

8-bromo-2-naphthol (43). A cold solution of NaNO\(_2\) (238.4 mg, 3.46 mmol) in water (2 ml) was added slowly to a solution of 8-amino-2-naphthol (500 mg, 3.14 mmol) in 48 % HBr (2.5 ml) and water (1.5 ml) at 0 °C and the reaction mixture was stirred 2 hrs at 0 °C. When diazotization was complete, CuBr (450.4 mg, 3.14 mmol) was added and suspension was heated at 70 °C for 1 hr. The reaction mixture was basified with 1 N NaOH and ethyl acetate was added. The organic layer was dried over anhydrous MgSO\(_4\), filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (3 : 1) as the eluent to obtain the product (33.4 %, white solid). \( ^1 \text{H} \) – NMR (400 MHz,
CDCl₃ δ 7.75-7.72 (m, 3H), 7.55 (d, J = 2.3 Hz, 1H), 7.19-7.13 (m, 2H), 5.26 (s, 1H)

8-bromo-2-naphthylamine (2p). Following the procedure of 2m, the compound was prepared (96.2 %, khaki solid): ¹H – NMR (400 MHz, CDCl₃) δ 7.68-7.63 (m, 3H), 7.35 (d, J = 2.2 Hz, 1H), 7.06 (dd, J = 8.0, 7.5 Hz, 1H), 6.96 (ddd, J = 8.7, 2.3, 2.3 Hz, 1H), 4.00 (bs, 2H)

1-(acetylamino)-4-bromonaphthalene (45). Acetic anhydride (0.48 ml, 4.95 mmol) was added dropwise to a solution of 4-bromonaphthalene (1 g, 4.5 mmol) in MeOH (8 ml) at 0 °C for 1 hr. The reaction mixture was refluxed for 1 hrs. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO₄, filtered and concentrated in vacuo to obtain product (95.8 %, skyblue solid). ¹H – NMR (400 MHz, CDCl₃) δ 8.28 (d, J = 8.1 Hz, 1H), 7.85 (d, J = 7.9 Hz, 1H), 7.77 (d, J = 8.2 Hz, 1H), 7.64-7.58 (m, 1H), 7.58 (bs, 1H), 2.33 (s, 3H)

1-(acetylamino)-4-bromo-2-nitronaphthalene (46). Fuming nitric acid (0.12 ml, 2.71 mmol) was added dropwise to a solution of 1-(acetyl amino)-4-bromonaphthalene (650 mg, 2.46 mmol) in acetic acid (7 ml) at 45 °C. The reaction mixture was heated for at 75 °C 30 min. The suspension was washed with cold MeOH and diethyl ether. The product (70.4 %, yellow solid) was filtered and dried in vacuo. ¹H – NMR (400 MHz, CDCl₃) δ 8.72 (bs, 1H),
8.38 (s, 1H), 8.30 (d, $J = 8.4$ Hz, 1H), 8.06 (d, $J = 8.5$ Hz, 1H), 7.80 (dd, $J = 7.6$, 7.6 Hz, 1H), 7.69 (dd, $J = 7.6$, 7.2 Hz, 1H), 2.37 (s, 3H)

4-bromo-2-nitro-1-naphthylamine (47). A solution of 1-(acetylamo)-4-bromo-2-nitronaphthalene (535 mg, 1.73 mmol) in 50 % H$_2$SO$_4$ (10 ml) and EtOH (10 ml) was stirred at ambient temperature for 4hrs. The suspension was washed with diethyl ether. The product (93 %, orange solid) was filtered and dried in vacuo. $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 8.47 (s, 1H), 8.23 (d, $J = 8.4$ Hz, 1H), 7.99 (d, $J = 8.4$ Hz, 1H), 7.78 (dd, $J = 8.2$, 7.2 Hz, 1H), 7.65 (dd, $J = 8.1$, 7.3 Hz, 1H), 7.37 (bs, 2H)

4-bromo-2-nitronaphthalene (48). Acetic acid (3 ml) was added slowly to a mixture of 4-bromo-2-nitro-1-naphthylamine (400 mg, 1.5 mmol) and NaNO$_2$ (175.7 mg, 2.55 mmol) in con. H$_2$SO$_4$ (3 ml) at 10-20 °C. The reaction mixture was stirred 2 hrs at 10-20 °C. When diazotization was complete, the reaction mixture and EtOH (3 ml) was added to a solution of cupric sulfate (274.7 mg, 1.65 mmol) in 32 % hypophosphorous acid (3 ml) at 10 ~ 20 °C. The suspension was heated at 55 °C for 1 hr. The reaction mixture was basified with 1 N NaOH and ethyl acetate was added. The organic layer was dried over anhydrous MgSO$_4$, filtered and concentrated in vacuo. The residue was purified by column chromatography on silica gel with nhexane/ethyl acetate (3 : 1) as the eluent to obtain the product (65.4 %, brown solid). $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 8.78 (d, $J = 1.9$ Hz, 1H), 8.57 (d, $J = 2.2$ Hz, 1H), 8.33 (d, $J = 8.5$ Hz, 1H), 8.06 (d, $J = 8.2$ Hz, 1H), 7.85-7.80 (m, 1H), 7.73-7.69 (m, 1H)

4-bromo-2-naphthylamine (2q). A mixture of 4-bromo-2-nitronaphthalene (240 mg, 0.95 mmol) dissolved in EtOH (3 ml) and tin (II) chloride dihydrate (2.1 g, 9.5 mmol) was refluxed for 3 hrs. The solvent was evaporated and the residue was poured into water and extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO$_4$, filtered and concentrated in vacuo to obtain product (79 %, brown solid). $^1$H – NMR (400 MHz, CDCl$_3$) $\delta$ 8.05 (d, $J = 8.4$ Hz, 1H), 7.56 (d, $J = 8.2$ Hz, 1H), 7.39 (ddd, $J = 8.2$, 6.8, 1.2 Hz, 1H), 7.33-
7.29 (m, 1H), 7.29 (d, \( J = 2.2 \) Hz, 1H), 6.95 (d, \( J = 2.1 \) Hz, 1H), 3.83 (bs, 2H)